

High performance OPV and their application in wearable devices

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Outline

The Status and Challenges of OPV

CH series of high performance **OPV** materials

Wearable OPV devices

The OPV status

PCE: $\sim 20\%$, still significantly behind of that of inorganic ones

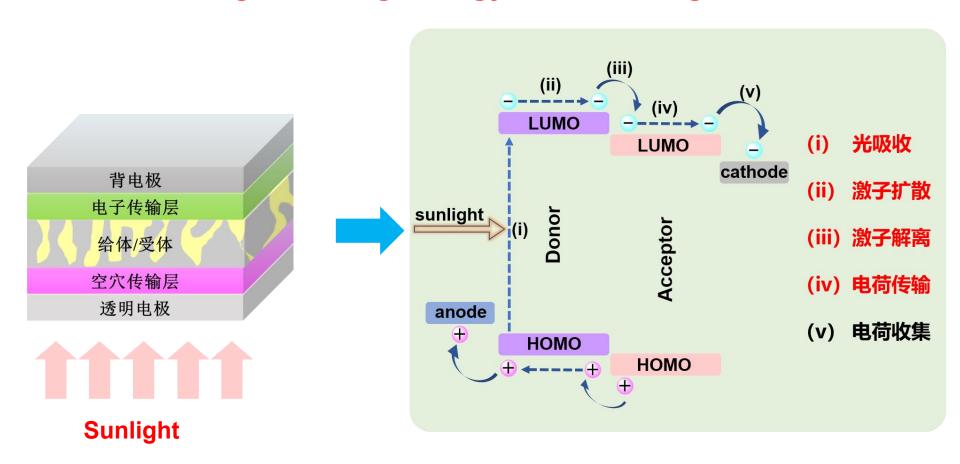
Stability issue

Application cases or where/what to use them?

Current issue for OPV

Materials in the active layer!

The issue: large binding energy leads to large Eloss!

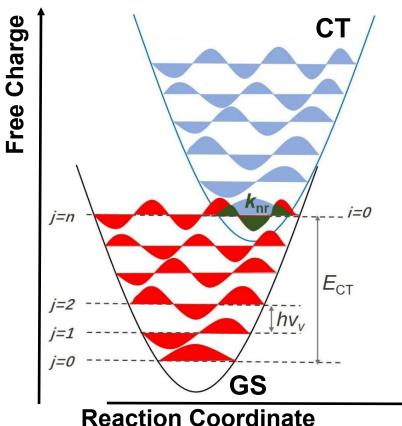


□ Most important process/steps happen at the interface!

Why for the large OPV Eloss?

Reason for large Eloss: serve no-radiatve loss

Mulliken-Hush Model



J. Am. Chem. Soc. 1952, 74, 811. Electrochim. Acta 1968, 13, 1005. How to reduce: E_{loss}

OPVFrenkel Exciton

Si/PVK
Wannier Exciton

- no-radiative loss
 GaAs/Si ~ 0.04 eV
 OPV > 0.16 eV
- Material optimization for reduced Krn!

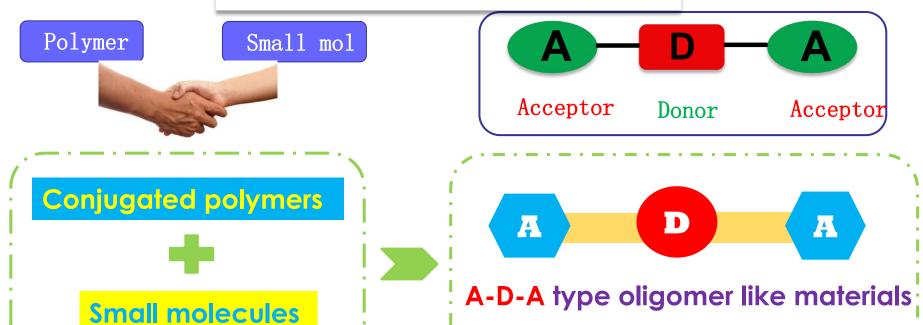
All matters:

- ground/excited/CT states
- Intermolecular interaction
- packing, morphology, etc

Some of our earlier works

High performance A-D-A molecules!

defined structures/molecules



Nature Electronics, 2019, 2, 513 Science 2018, 361, 109 Nature. Photon. 2017, 11, 85 Nature Photon., 2015, 9, 35 Acc. Chem. Res., 2013, 46, 2645 Adv. Mater, 2019,1804723
Adv. Mater, 2018, 30, 1707508
J. Am. Chem. Soc, 2015, 137, 3886
J. Am. Chem. Soc, 2014, 136, 15529
J. Am. Chem. Soc., 2013, 135, 8484
J. Am. Chem. Soc., 2013, 135, 5921
J. Am. Chem. Soc., 2012, 134, 16345

J. Am. Chem. Soc., 2013, 135, 5921



A semi-empirical model for OPV

SQ (Shockley-Queisser) PV model, Alan Heeger OPV model

A semi-empirical model for OPV

$$V_{oc} = \frac{1}{q} (E_g - E_{loss}) = \frac{1}{q} (\frac{1240}{\lambda} - E_{loss})$$

$$J_{sc} = \int_{300}^{\lambda} \frac{q\lambda}{hc} \cdot E(\lambda) \cdot EQE(\lambda) \cdot d\lambda$$

$$PCE(\%) = V_{oc} \cdot J_{sc} \cdot FF/P_{in} = \frac{1}{e} \left[\left(\frac{1240}{\lambda} - E_{loss} \right) \cdot \int_{300}^{\lambda} \frac{q\lambda}{hc} \cdot E(\lambda) \cdot EQE(\lambda) \cdot d\lambda \cdot FF/P_{in} \right]$$

Based on

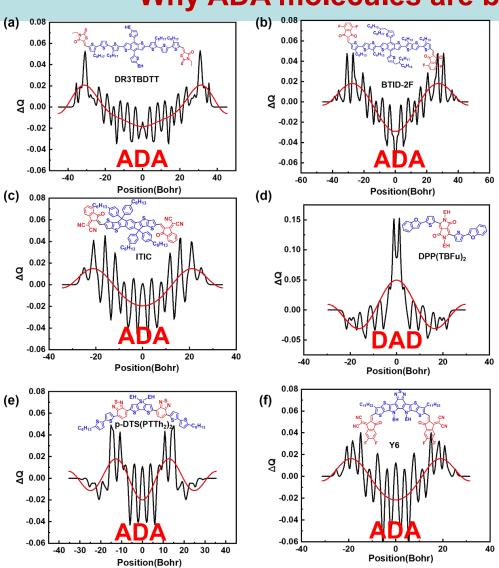
- Balanced Theory
- Shockley-Queisser Limit Theory
- (SQ Limit for photovoltaic device)
- State-of-the-art experimental results



A semi-empirical model, for both material design and optimiza

The characteristics of ADA type compounds

Why ADA molecules are better: A-D-A vs D-A-D



What are ADA molecules (vs DAD)? most unique characteristics/特征?

HOMO/LOMO Electron density distribution

$$\Delta \mathbf{Q} = \mathbf{\Psi}_{LUMO}^2 - \mathbf{\Psi}_{HOMO}^2$$

Charge density difference between excited state and ground state

U vs reverse T type frontier electron density distribution

The characteristics of ADA materials

Why A-D-A is better than DAD etc?

Unique and optimal

spatial electron distribution at front (excited) orbitals



- 1) Facilitated exciton dissociation
- 2) Optimized Morphology for charge transportation
- 3) Stable and 2D/3D networked packing
- 4) Smaller Eloss



How we can get better?

Chem. Soc. Rev. 2020, 49, 2828.

Contents

The Status and Challenges of OPV

CH series of high performance **OPV** materials

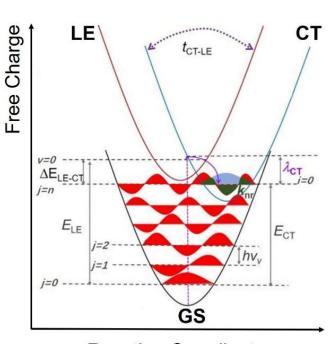
Wearable OPV devices

Charactristics of CH Series OPV molecules

2D conjugation extention for smaller reorganization energy and higher FL

molecules

非富勒烯体系"三态" 模型



Reaction Coordinate *Nat. Energy* 2021, *6*, 799.

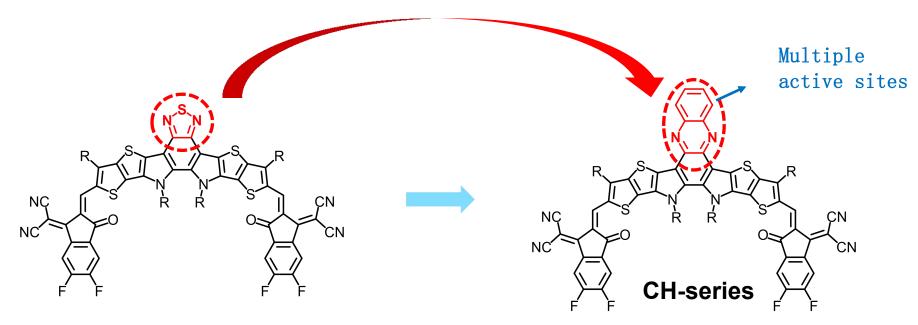
Get better molecules:



- ☐ High flurescence
- ☐ Increase CT state
- □ Reduce the coupling between CT & GS

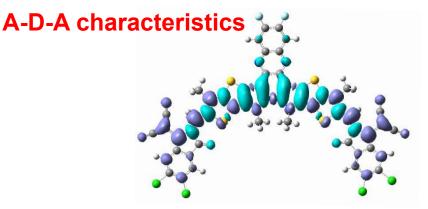
Some examples of CH series materials

1. Central unit conjugation extension



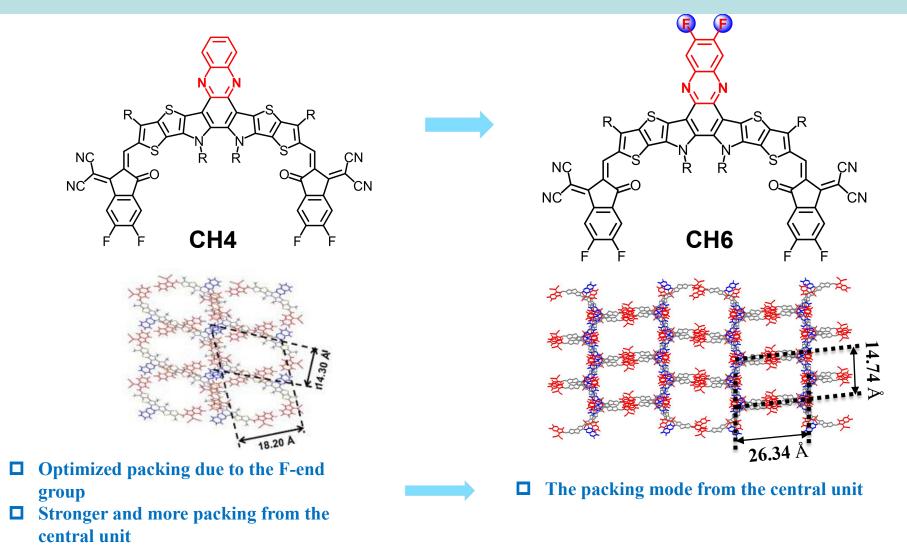
Advantages:

- □ smaller reorganization energy
- □ stronger intermol packing
- □ Optimze packing modes
- □ Better potential for optimization

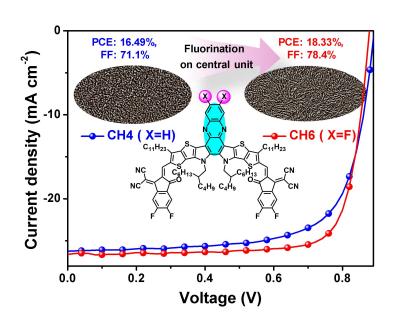


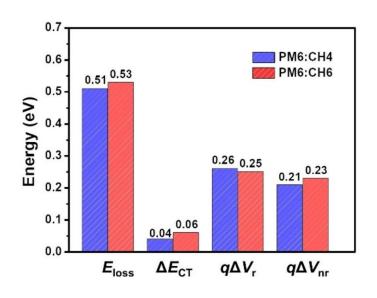
$$\Delta Q = \Psi_{LUMO}^2 - \Psi_{HOMO}^2$$

2. F-subsitituted central unit



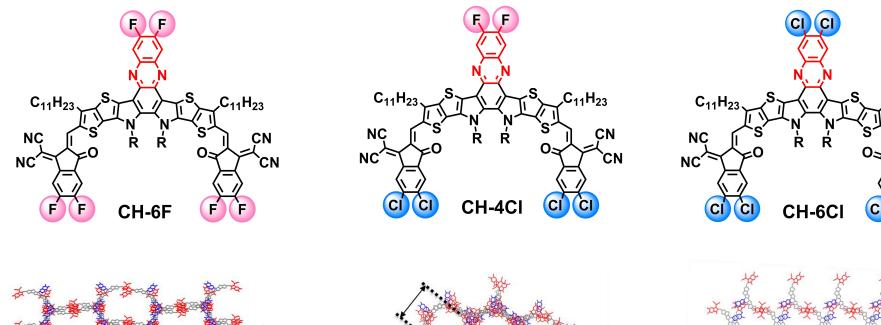
F-substitution leads to better performance





- ☐ Better morpholohy and higher PCE (18.33% vs 16.49%)
- **□** Smaller Eloss

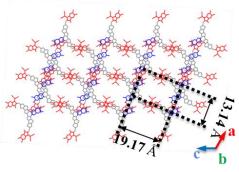
3. More peripheral X substitution: 18.22% PCE



CH-6F monoclinic system

c b—a

CH-4Cl trigonal system



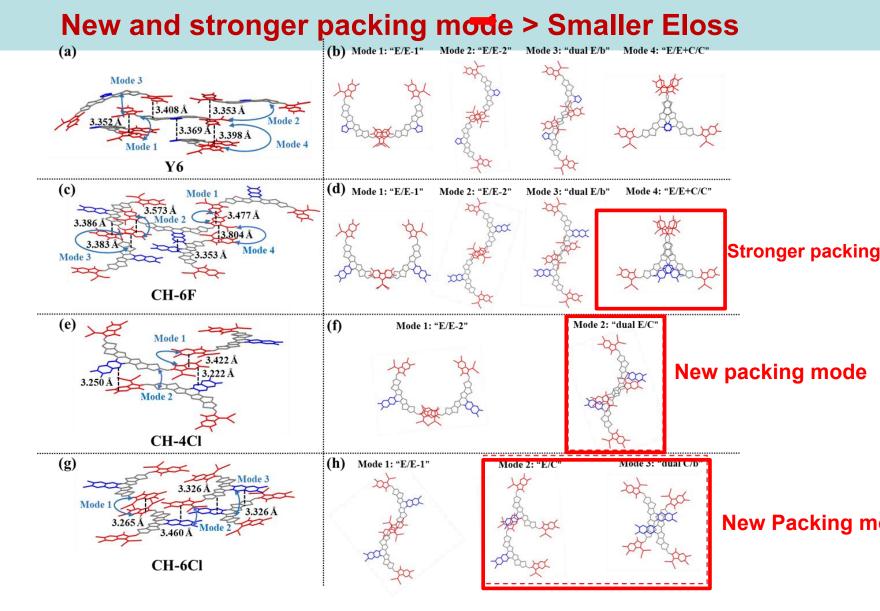
CH-6Cl triclinic system

☐ Fine tuning of X

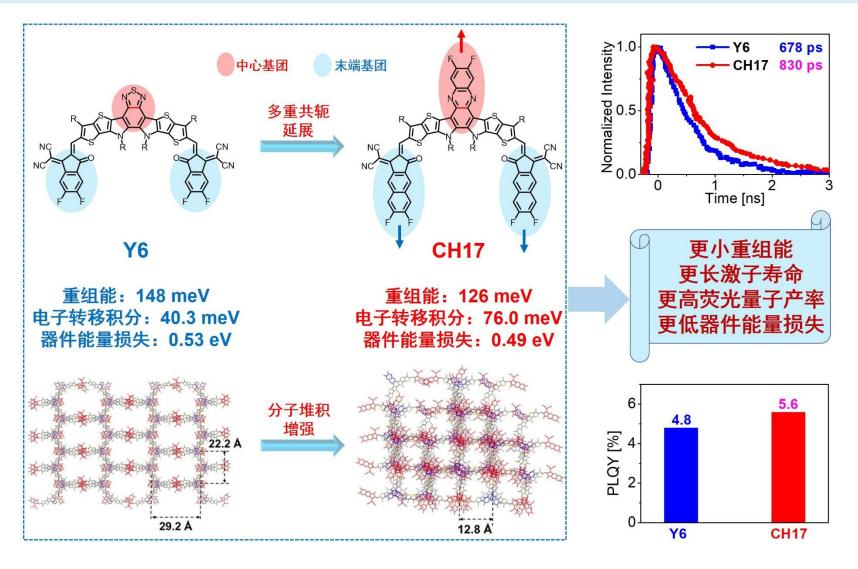


□ Packing changed singnificantly

Energy Environ. Sci. 2022, 15, 3519 Angew. Chem. Int. Ed., 2023, 61, e202312630. C₁₁H₂₃

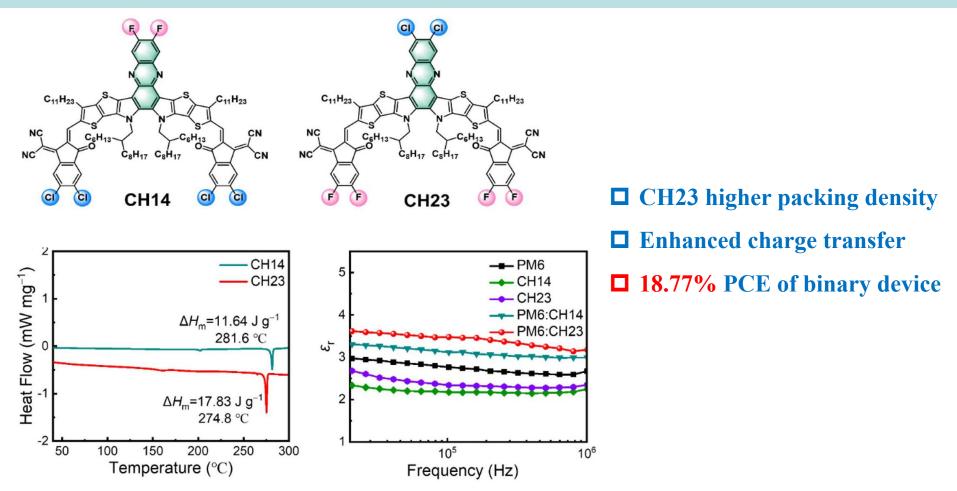


4. Mutiple conjugation extention



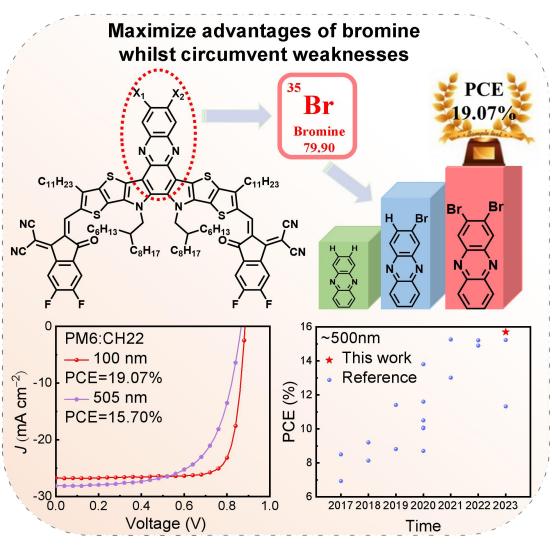
Sci. China Chem. 2022, 65, 1362.

5. Dielectric consistant turning



larger dielectric constant due to enhance packing & halogen subsitition

Bromided CH material for PCE > 19%



Br atom and related molecules

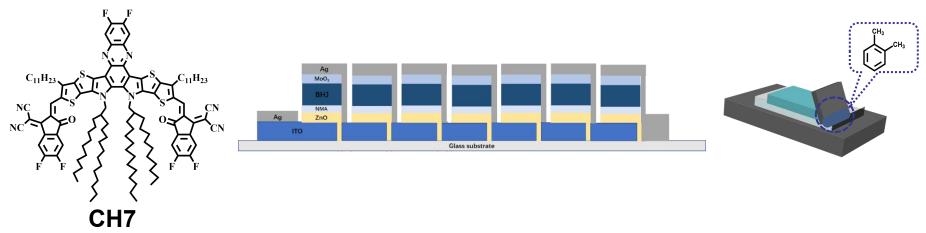
- Easier polerization, better crystalline
- **□** Larger steric hindance
- ☐ Optimized approach: introduce

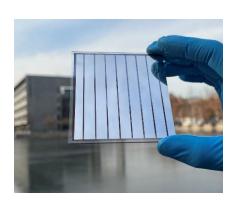
 Br at the central unit
- **□** Stronger and more ordered packing
- □ Larger E、Smaller E of binding
- □ PCE of 19.06% for binary device
- □ 500 nm thickness device with 15.7%PCE

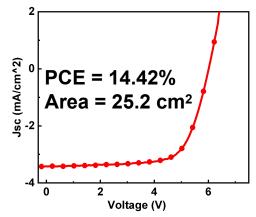
A race of for brmoide materials

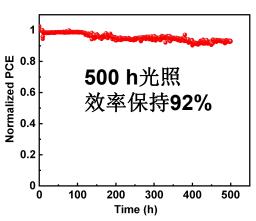
6. OPV module

- **♦** Blade coating for large OPV module (25 cm²)
- ◆ PCE of 14.42%, 92% remains after 500 h under MPP



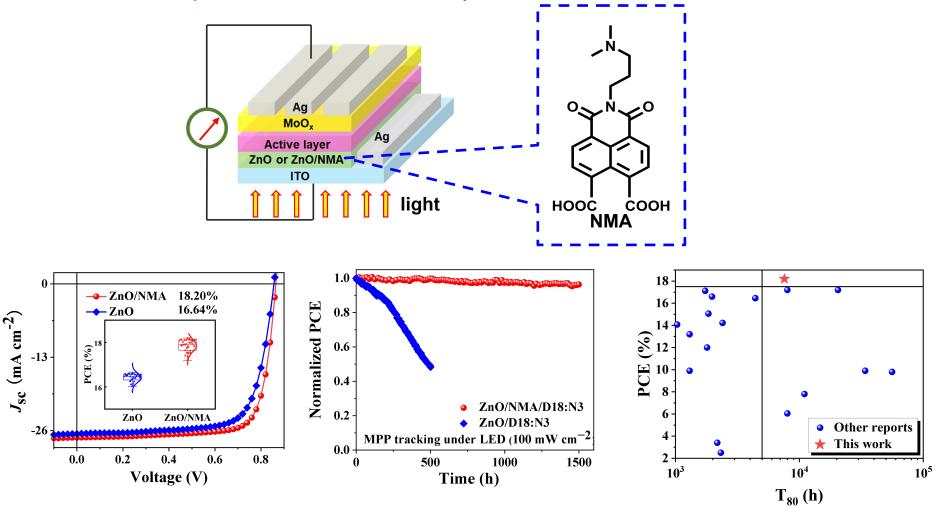






Both stable and high performance OPV

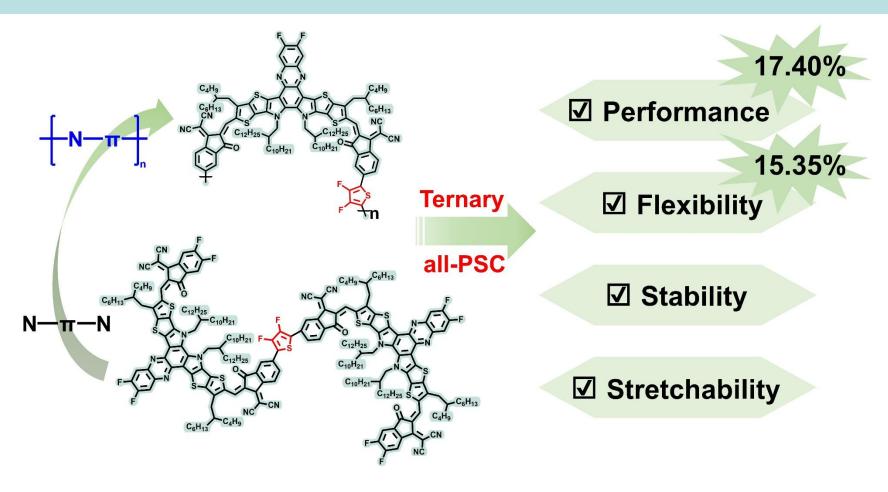
- **◆** HTL NMA—High reverse OPV with high performance
- ◆ Life time > 5 years—The best stability with PCE > 18% under MPP conditions



Angew. Chem. Int. Ed.., 2022, e202207397

Dimer of CH acceptor materials

7. Dimers of CH series materials



Adv. Energy Mater. 2023, DOI: 10.1002/aenm.202300301.

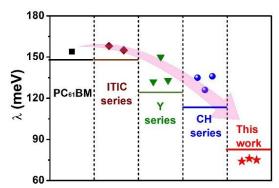
8. Dimers of CH molecules connected at the central unit

3D acceptors with multiple CH A-D-A units

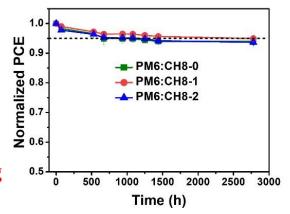
Monomer to Dimer Multiple Various modification **Bridge** building sites blocks **Novel dimerzation** Avoid interfered end packing

T_g increased and remained end packing

Reduced reorganization energy



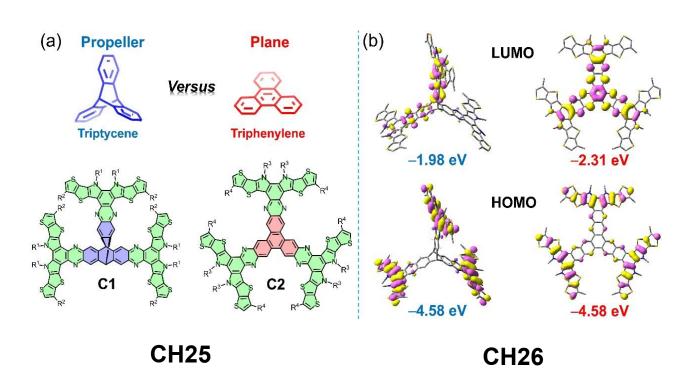
- $\square PCE > 17\%$
- **□** Enhanced stability



Energy Environ. Sci. 2023, 16, 1773

Angew. Chem. Int. Ed., 2023, e202307962

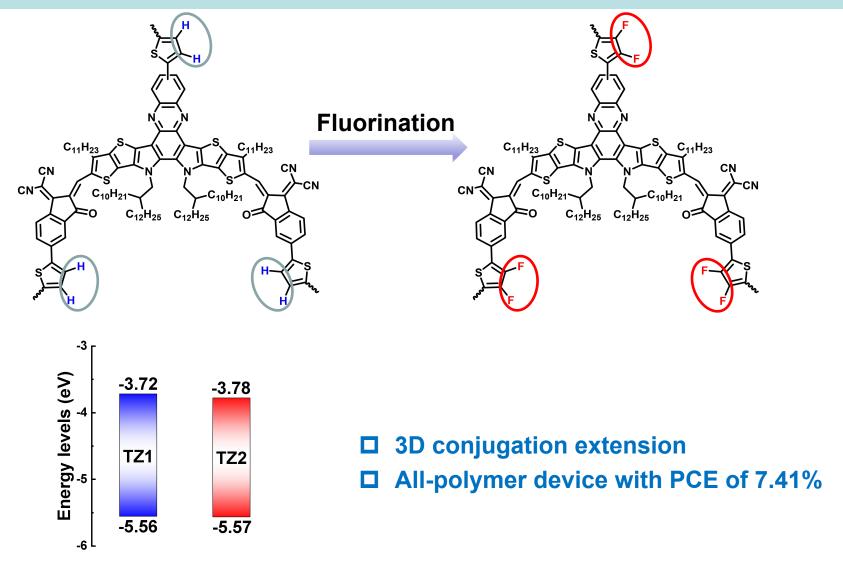
planar vs propeller type molecules



Active Layers	V _{oc} [V]	J _{SC} [mA cm ⁻²]	Calc. J _{SC} ^[b] [mA cm ⁻²]	FF [%]	PCE [%]
PM6:CH25	0.840 (0.836±0.010)	4.82 (4.19±0.35)	4.33	55.4 (55.4±0.8)	2.24 (1.91±0.19)
PM6:CH26	0.920 (0.922±0.003)	22.98 (22.60±0.22)	22.49	72.7 (77.3±0.4)	15.41 (15.11±0.17)

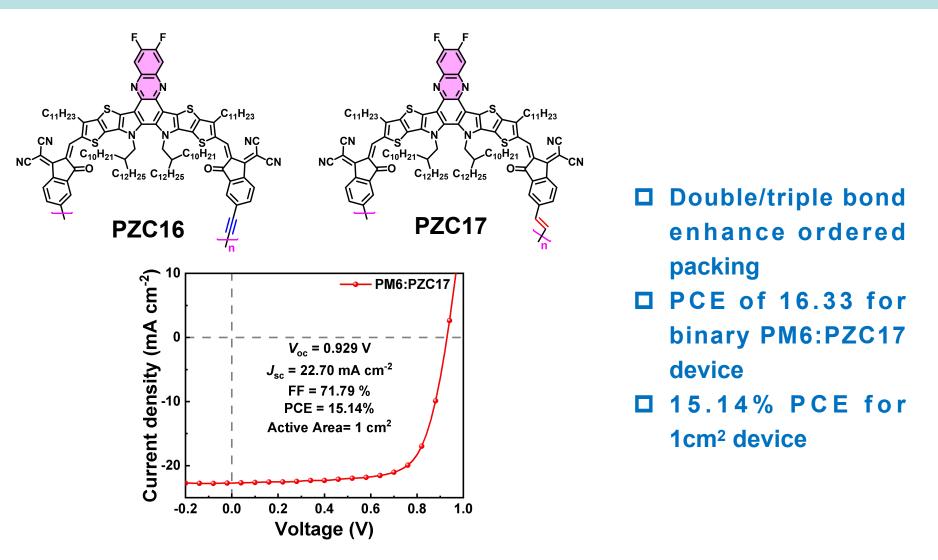
Polymerized CH acceptor materials

9. Polymerized CH type molecules



Org. Electron. 2023, *114*, 106735.

Different linker: double vs triple CC bonds



Adv. Funct. Mater. 2023, 2214248.

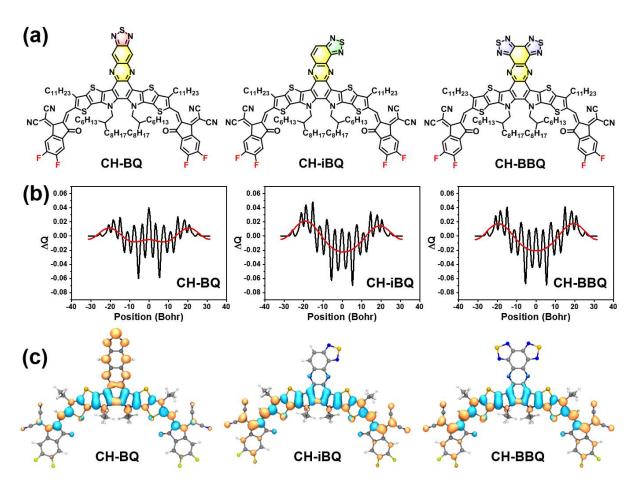
Short summary: CH is a great platform for multiple optimization

More diverised and optimization in terms of molecular design!

Single molecules, oligomers and polymers

More structure optimization = Better PCE imporvement possibility

Mutiple conjugation extension



PCE ~ 19%

Angew. Chem. Int. Ed., 2023, e202308832

Ontline

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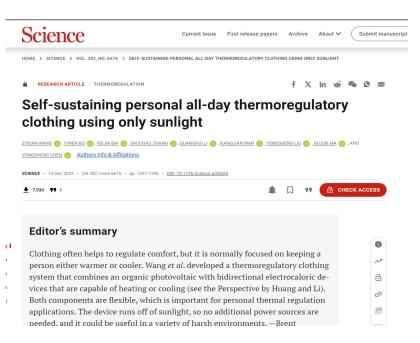
Where are OPV for?

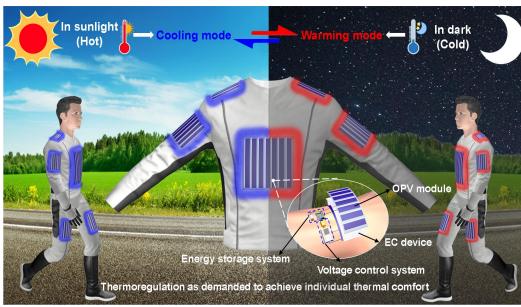
轻,薄,柔,透光 Light, thin, flexibe, transparent

Weable OPV and its applications

Solar-powered clothes, for the heat and cold

Flexible OPV-EC thermoregulatory clothing (OETC): cooling in sunlight, warming in dark

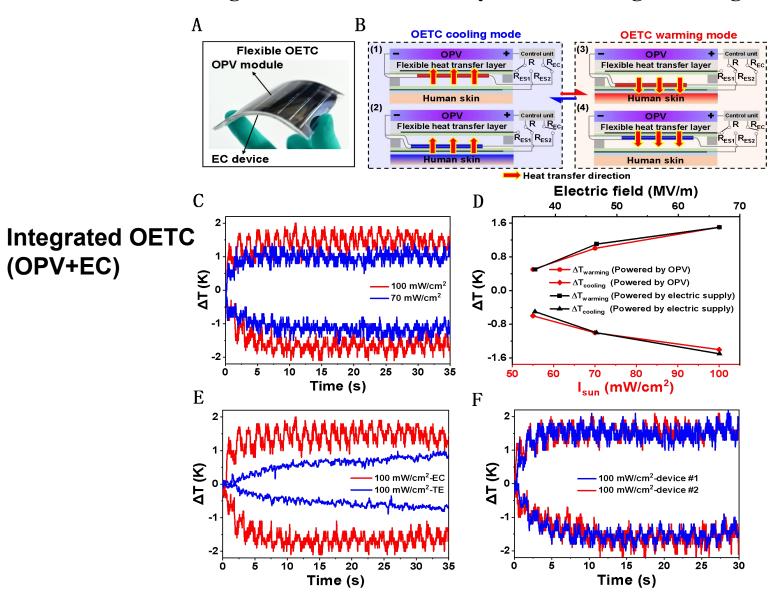




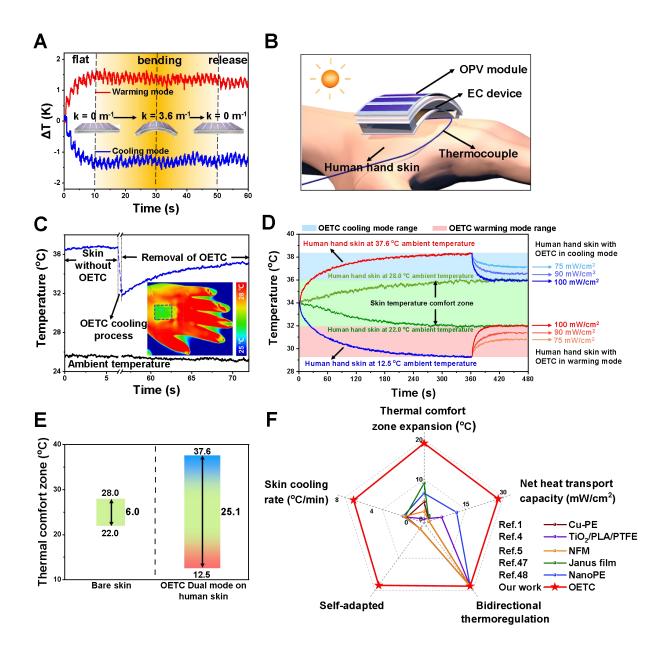
Outdoor, space, polar regions

Self-sustaining personal all-day thermoregulatory clothing using only sunlight

Working mechanism of OETC system in cooling/warming mode



Self-sustaining personal all-day thermoregulatory clothing using only sunlight

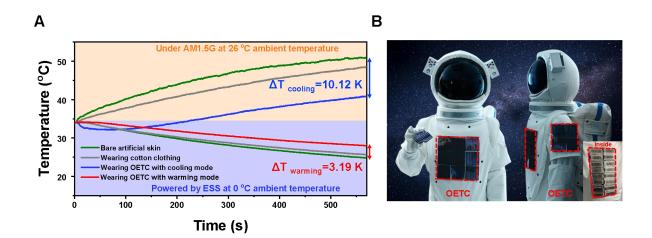


OPV+EC 集成器件

Self-sustaining personal all-day thermoregulatory clothing using only sunlight

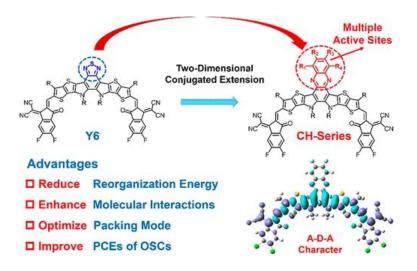
Thermoregulation performance of OETC in the outdoor and the prospect for use in space

OPV+EC 集成器件



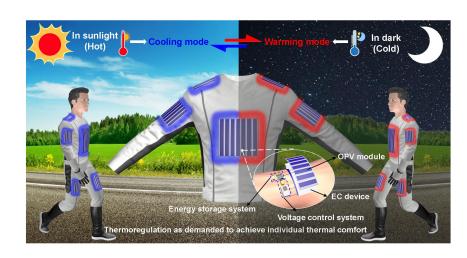
Summary

CH series of OPV molecules



Acc of Mater Res, **2023**, 4, 772 **Chem Soc Rev**, **2020**, 49 (9), 2828 **Acc Chem Res**, **2012**, 46, 2645

Flexible and wearable OPV devices



Cooling in sunlight, warming in dark!

Science, 2023, 328, 1291-1296

Acknowledgements



MoST, NSF, Tianjin City, Nankai Univ and many collablrators mostly, the students!

Comments and Discussion welcomed