



King Abdullah
University of Science
and Technology

A large, modern building with a glass facade and illuminated by its own lights, set against a dark sky. The building is reflected in the water in the foreground.

Organic Semiconductor Devices: from energetics to green processing

Derya Baran

Associate Professor of Materials Science and Engineering Program, KAUST
Sino-German Workshop ERLANGEN 2024

KAUST/ OMEGALAB

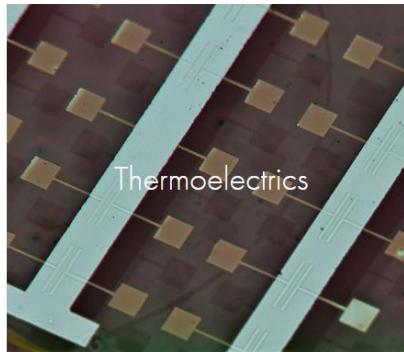




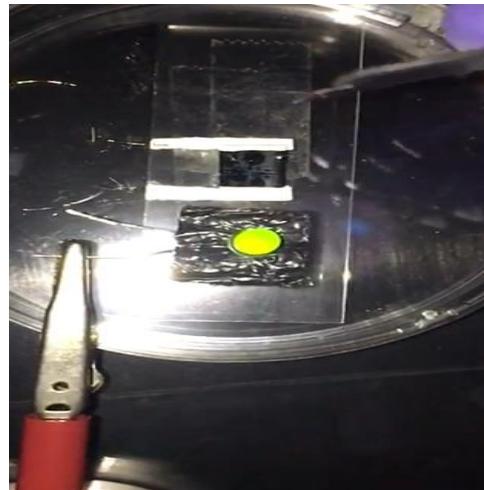
DERYA BARAN ORGANIC/HYBRID MATERIALS FOR ENERGY APPLICATIONS(www.omegalabresearch.com)



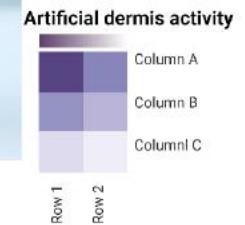
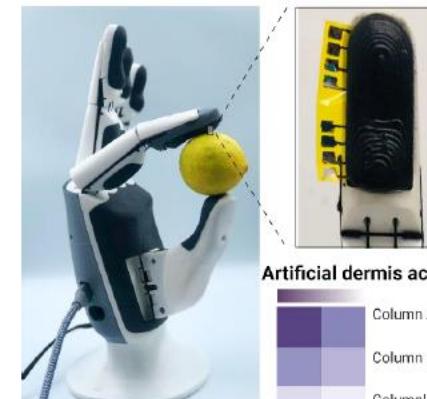
Solution processed PV
Stability and reliability of PV



Thermoelectrics
Charge transport in n-type
Stretchable healable TE



Printing, 3D printing, Micro printing
Conductive composites
Electronic Skin





جامعة الملك عبد الله
للغالوم والتكنولوجيا
King Abdullah University of
Science and Technology

KAUST
SOLAR
CENTER

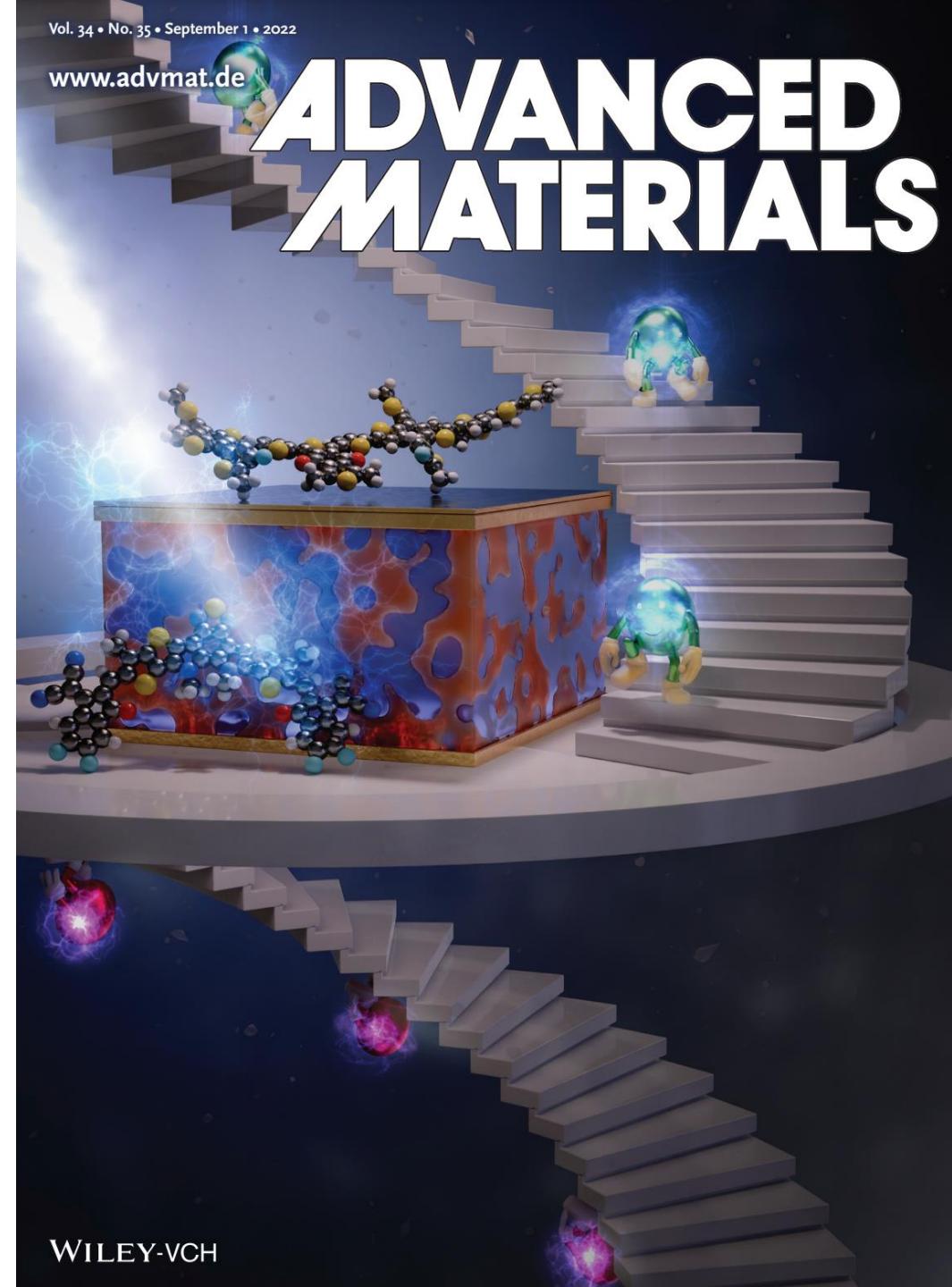
Energy level determination of organic semiconductors for solar cells



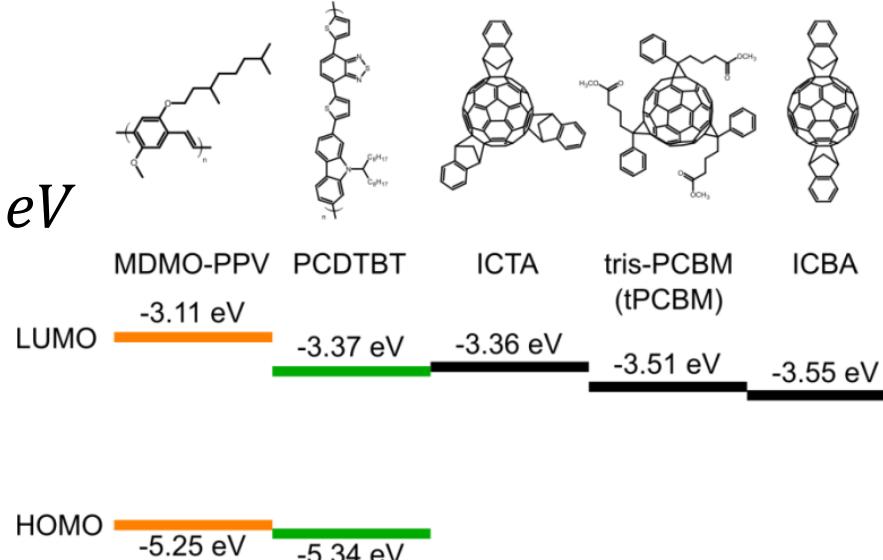
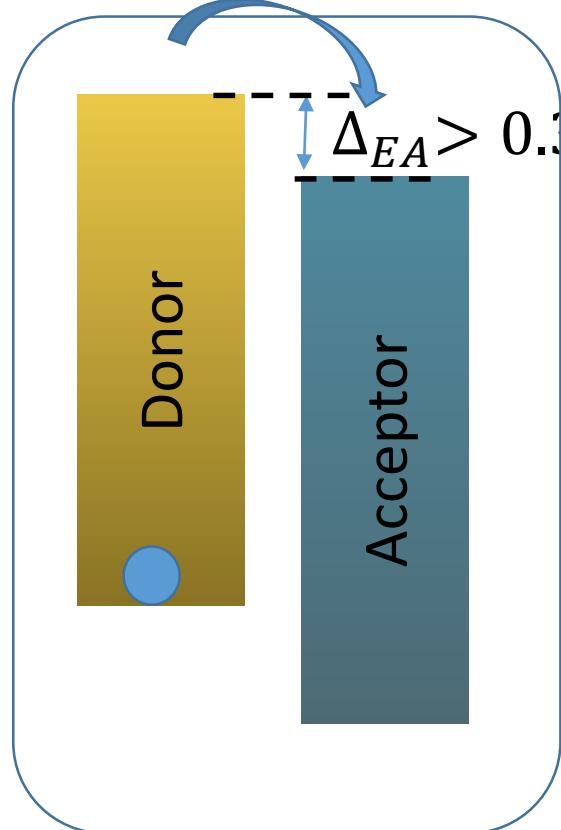
Anirudh Sharma



Jules Bertrandie

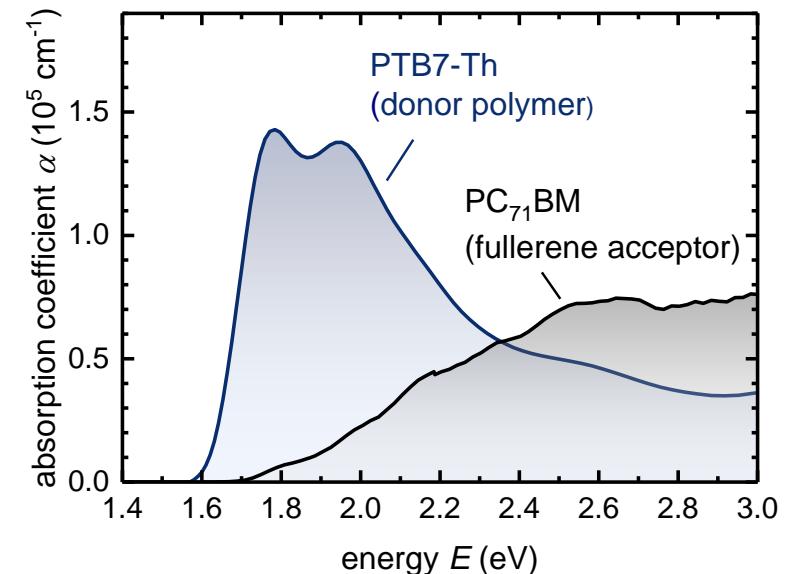


Why energetics matter?

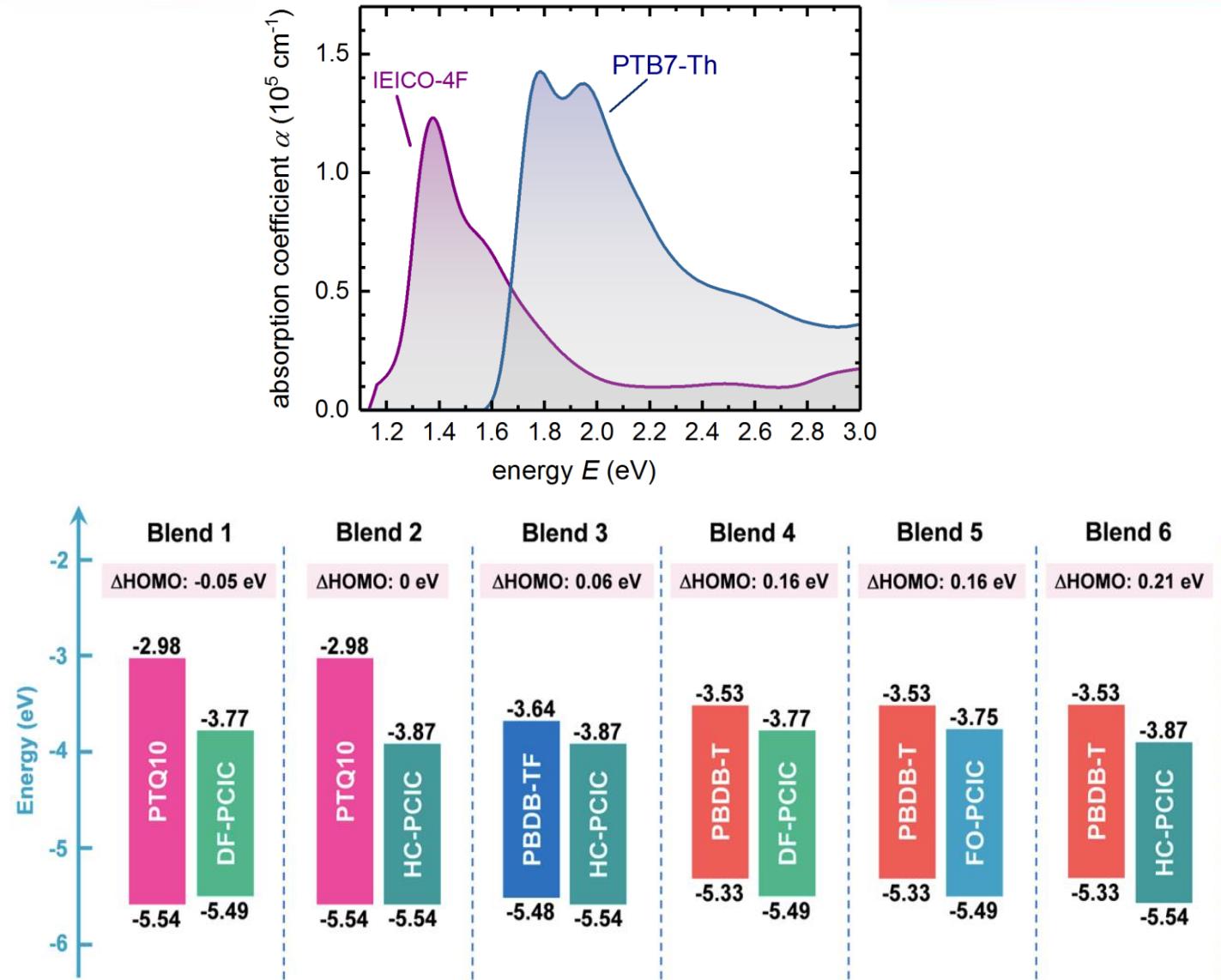
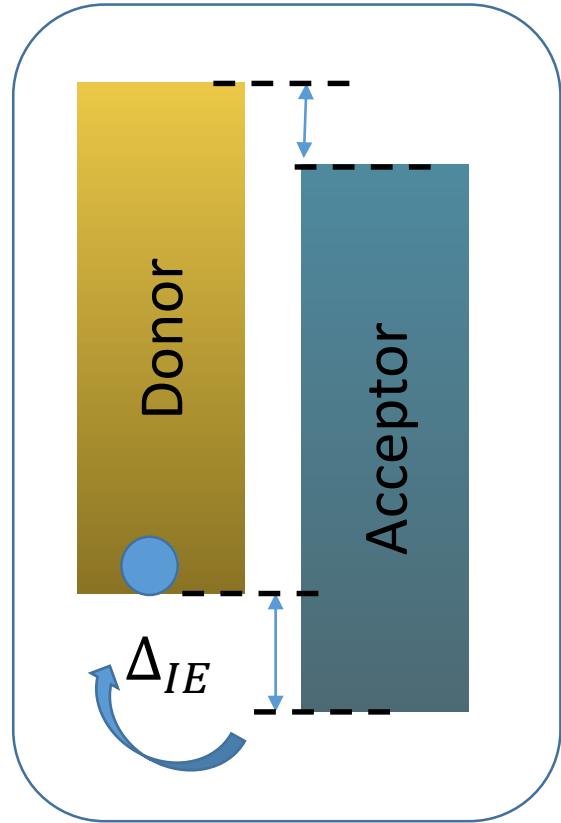


Very little driving energy is required at the heterojunction to dissociate excitons and generate free charges

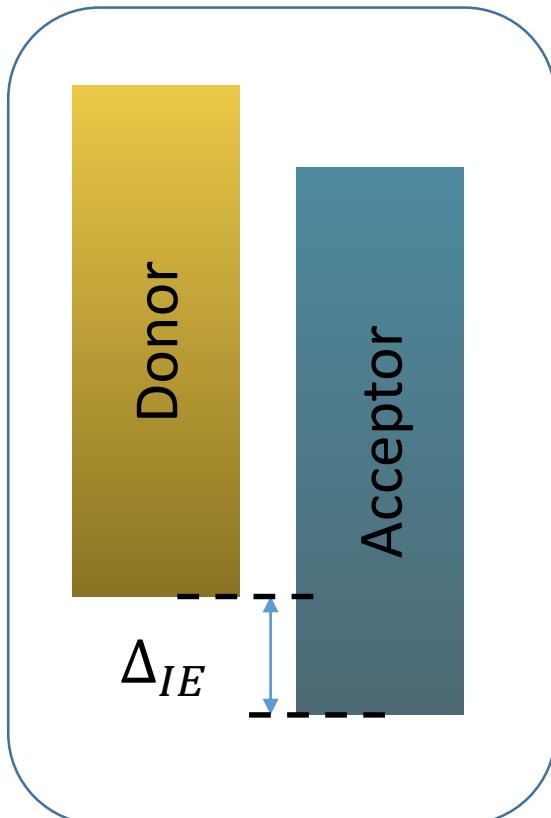
Fullerenes mostly had wider bandgap than donors



Why energetics matter?



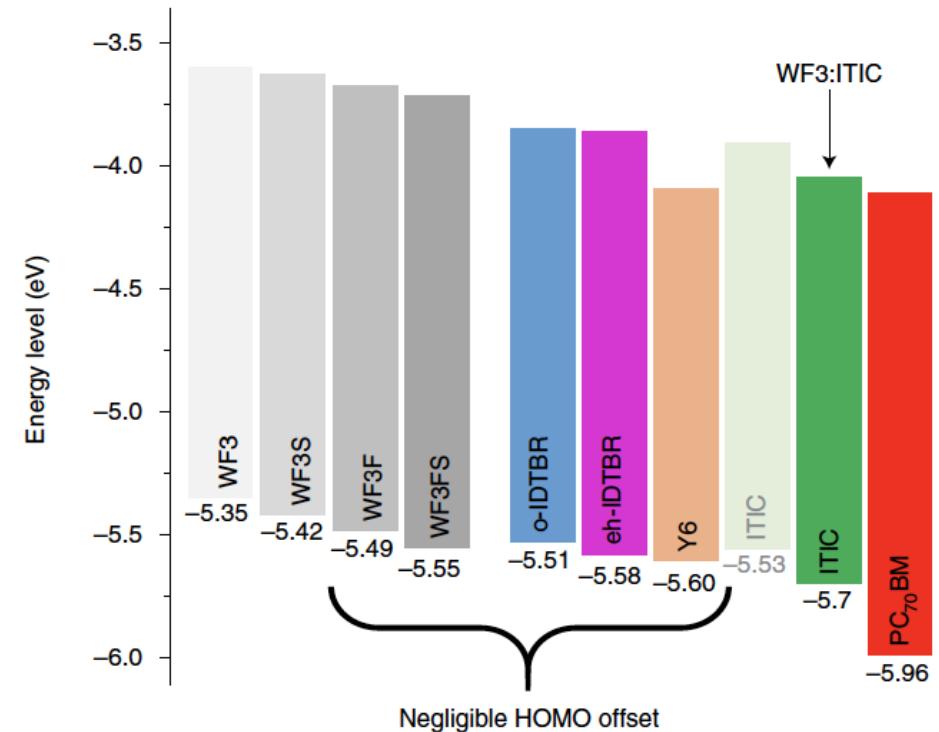
Offset or no offset?



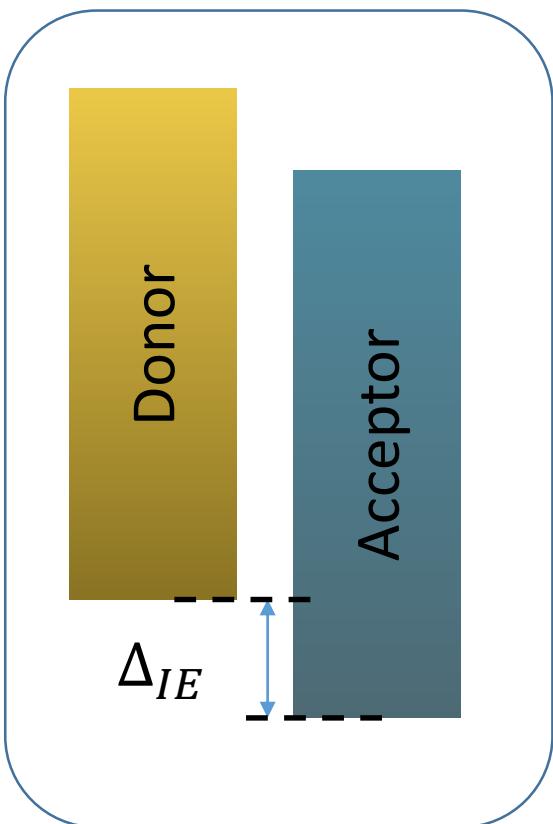
The role of exciton lifetime for charge generation in organic solar cells at negligible energy-level offsets

Andrej Classen^①, Christos L. Chochos^{2,3}, Larry Lüer^①✉, Vasilis G. Gregoriou^{2,4}, Jonas Wortmann^③, Andres Osset^④, Karen Forberich^④, Iain McCulloch^{⑤,6}, Thomas Heumüller^{③,7}✉ and Christoph J. Brabec^{①,7,8}✉

Organic solar cells utilize an energy-level offset to generate free charge carriers. Although a very small energy-level offset increases the open-circuit voltage, it remains unclear how exactly charge generation is affected. Here we investigate organic solar cell blends with highest occupied molecular orbital energy-level offsets (ΔE_{HOMO}) between the donor and acceptor that range from 0 to 300 meV. We demonstrate that exciton quenching at a negligible ΔE_{HOMO} takes place on timescales that approach the exciton lifetime of the pristine materials, which drastically limits the external quantum efficiency. We quantitatively describe this finding via the Boltzmann stationary-state equilibrium between charge-transfer states and excitons and further reveal a long exciton lifetime to be decisive in maintaining an efficient charge generation at a negligible ΔE_{HOMO} . Moreover, the Boltzmann equilibrium quantitatively describes the major reduction in non-radiative voltage losses at a very small ΔE_{HOMO} . Ultimately, highly luminescent near-infrared emitters with very long exciton lifetimes are suggested to enable highly efficient organic solar cells.



Offset or no offset?



ARTICLES

<https://doi.org/10.1038/s41563-020-00835-x>

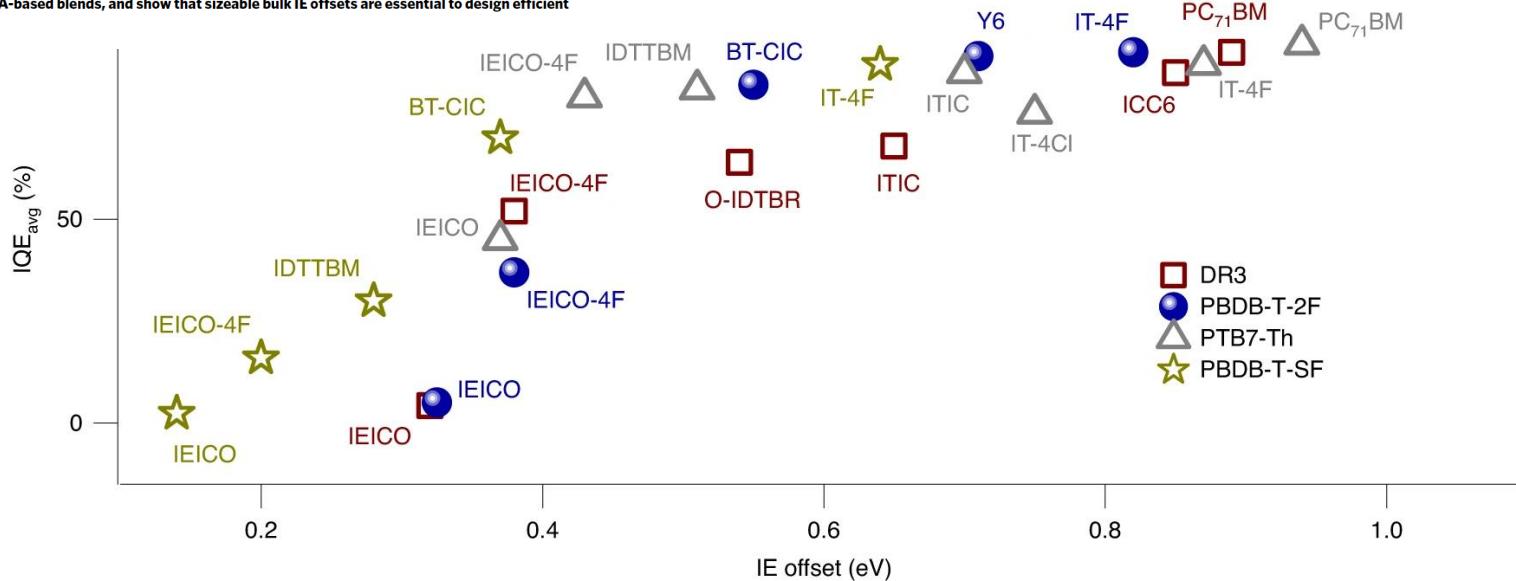
nature
materials

Check for updates

Intrinsic efficiency limits in low-bandgap non-fullerene acceptor organic solar cells

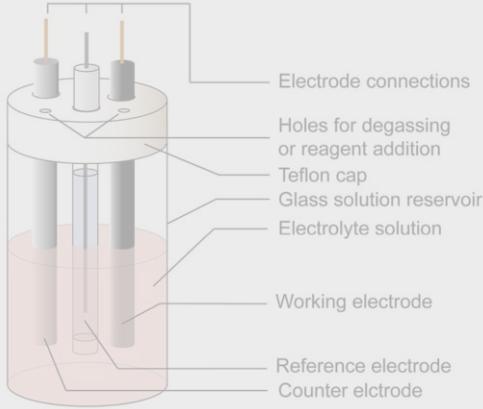
Safakath Karuthedath^{①,5}, Julien Gorenflo^{①,5}, Yuliar Firdaus^①, Neha Chaturvedi^①, Catherine S. P. De Castro^{①,5}, George T. Harrison^①, Jafar I. Khan^①, Anastasia Markina^②, Ahmed H. Balawi^①, Top Archie Dela Peña^①, Wenlan Liu^②, Ru-Ze Liang^①, Anirudh Sharma^①, Sri H. K. Paleti^①, Weimin Zhang^①, Yuanbao Lin^①, Erkki Alarousu^①, Dalaver H. Anjum^③, Pierre M. Beaujuge^①, Stefaan De Wolf^①, Iain McCulloch^{④,14}, Thomas D. Anthopoulos^①, Derya Baran^①, Denis Andrienko^② and Frédéric Laquai^①

In bulk heterojunction (BHJ) organic solar cells (OSCs) both the electron affinity (EA) and ionization energy (IE) offsets at the donor-acceptor interface should equally control exciton dissociation. Here, we demonstrate that in low-bandgap non-fullerene acceptor (NFA) BHJs ultrafast donor-to-acceptor energy transfer precedes hole transfer from the acceptor to the donor and thus renders the EA offset virtually unimportant. Moreover, sizeable bulk IE offsets of about 0.5 eV are needed for efficient charge transfer and high internal quantum efficiencies, since energy level bending at the donor-NFA interface caused by the acceptors' quadrupole moments prevents efficient exciton-to-charge-transfer state conversion at low IE offsets. The same bending, however, is the origin of the barrier-less charge transfer state to free charge conversion. Our results provide a comprehensive picture of the photophysics of NFA-based blends, and show that sizeable bulk IE offsets are essential to design efficient BHJ OSCs based on low-bandgap NFAs.

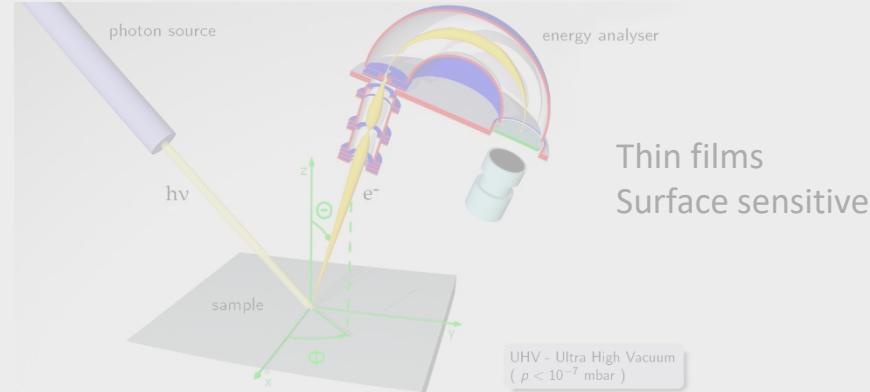




Motivation



Solutions; thin films
Presence of electrolyte
Electrochemical instability
Lack of electrochemical reversibility



Thin films
Surface sensitive

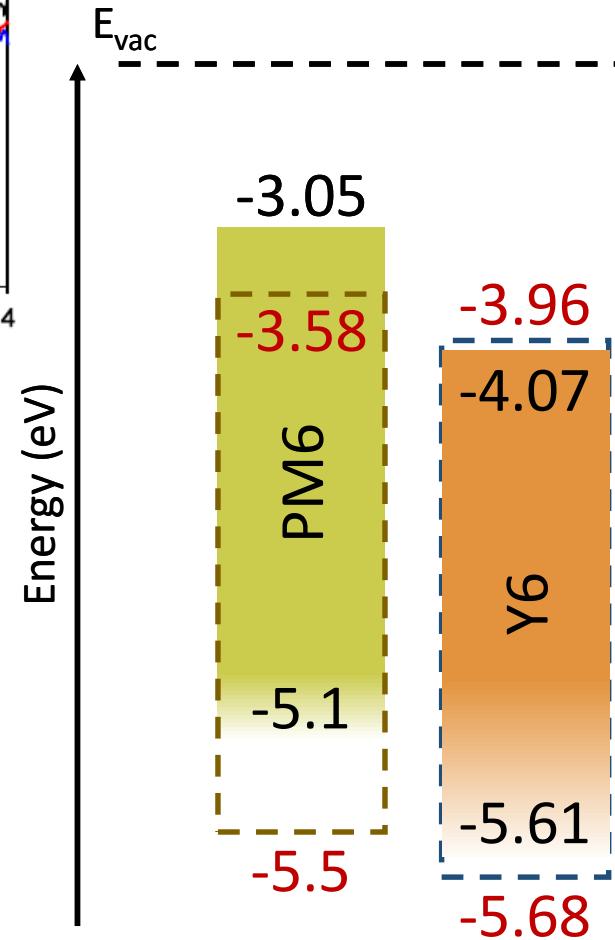
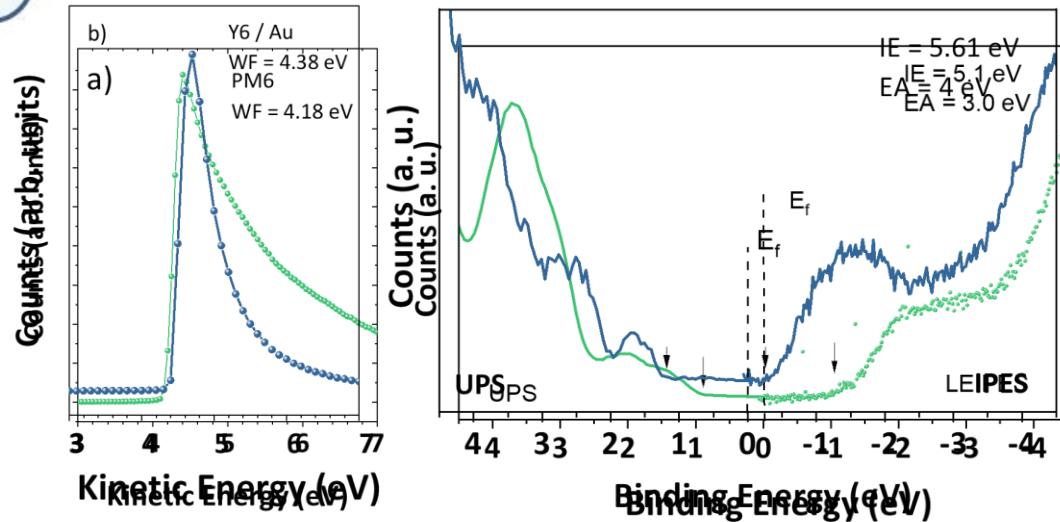
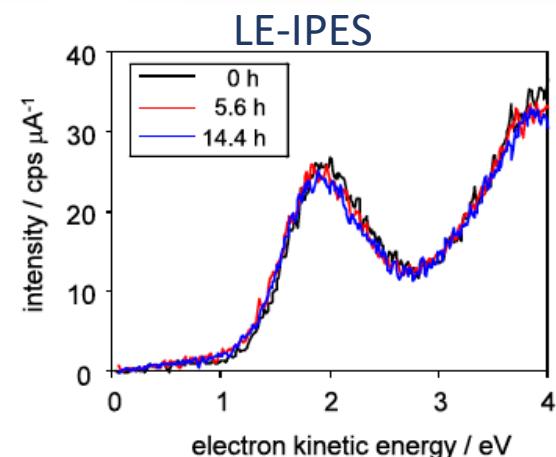
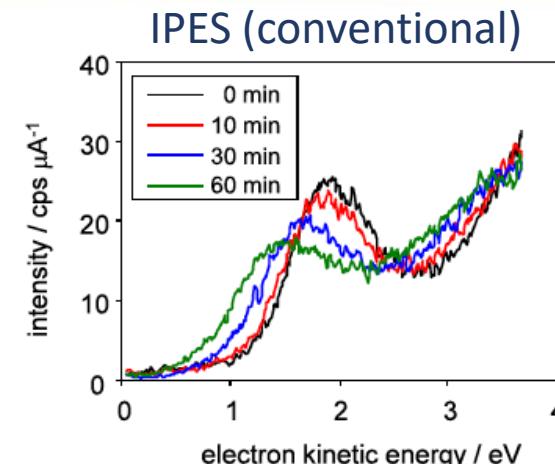
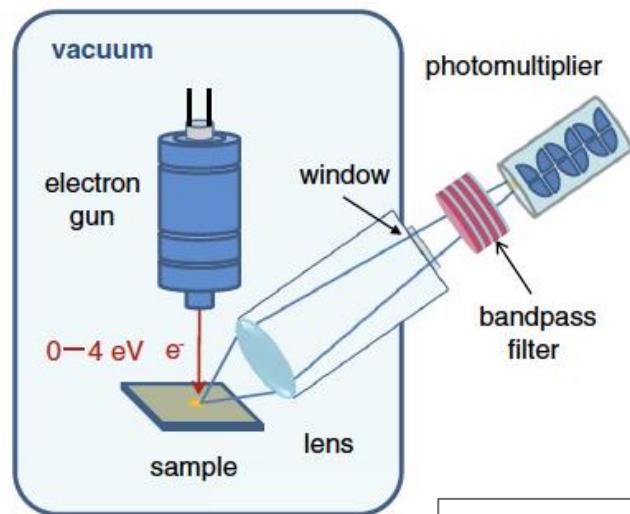
Define a criteria for determining the energetics of OSCs

Correlation between material energetics – Voc

Establish material-property relationship for design rules

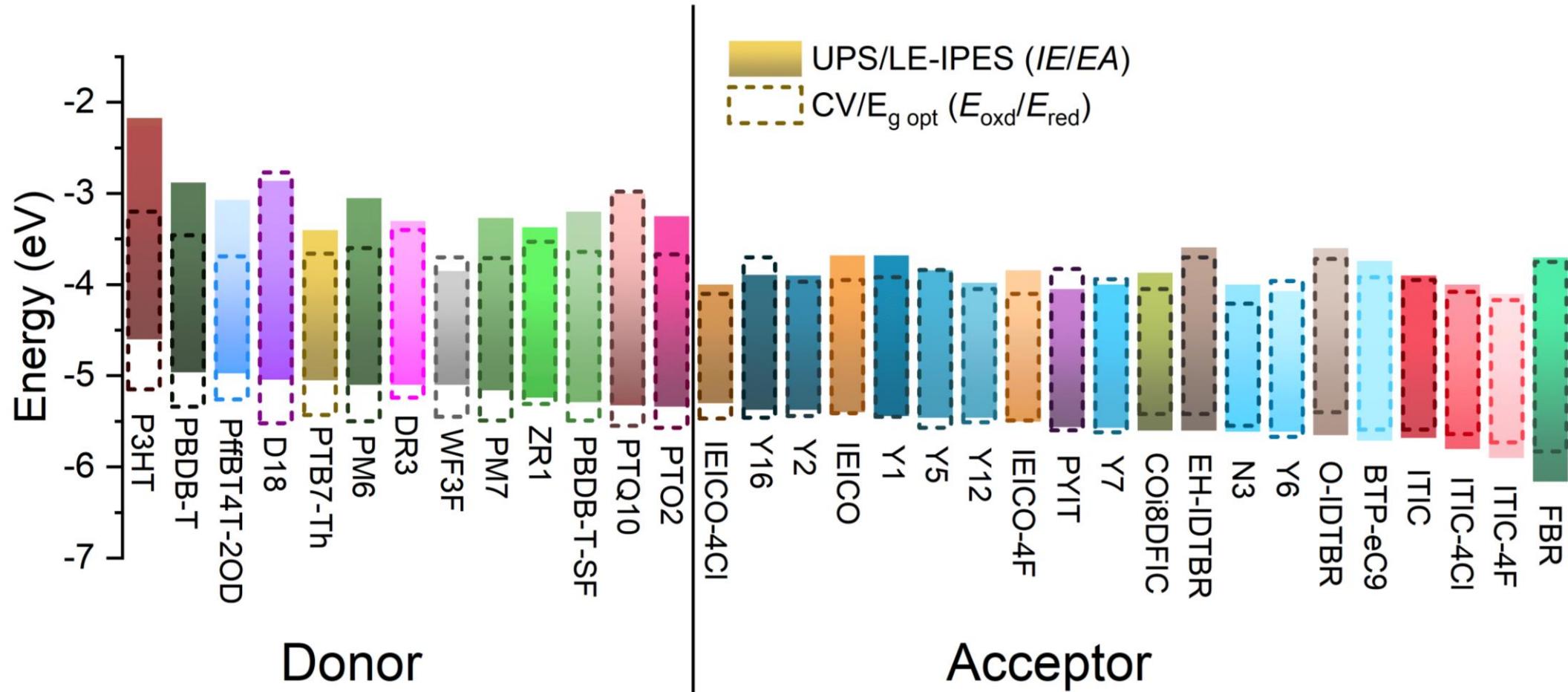


Method of measurements





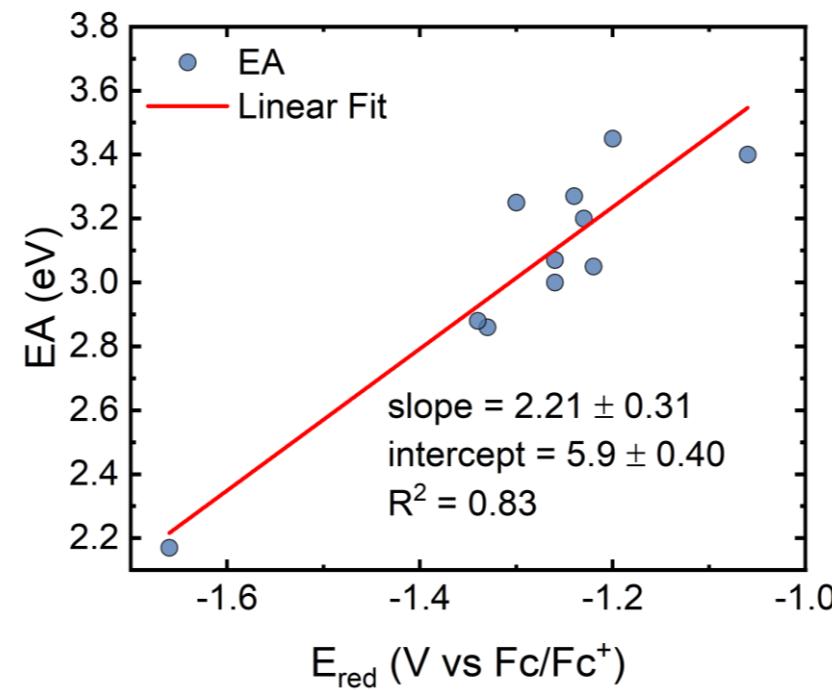
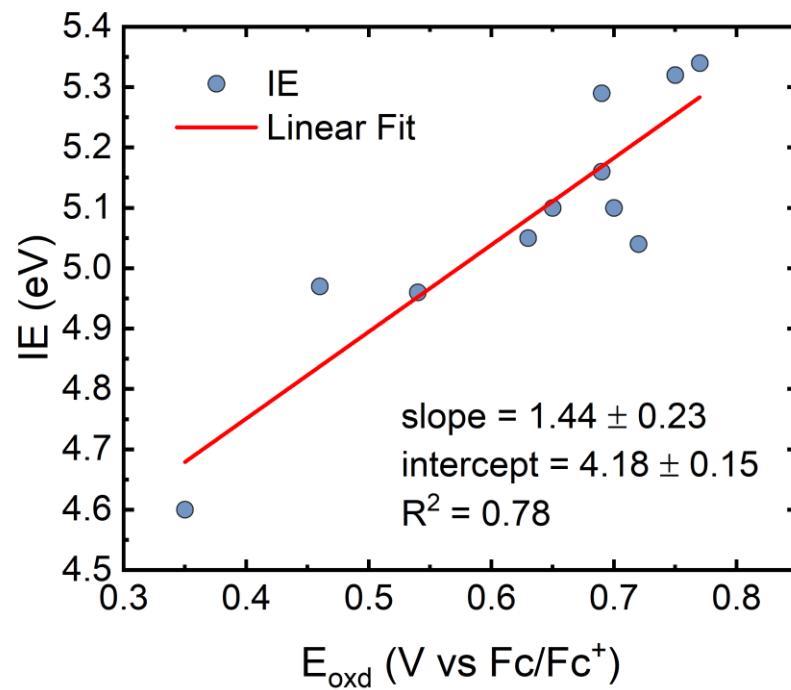
Library of Energetics





Correlating Redox Potentials with IE/EA

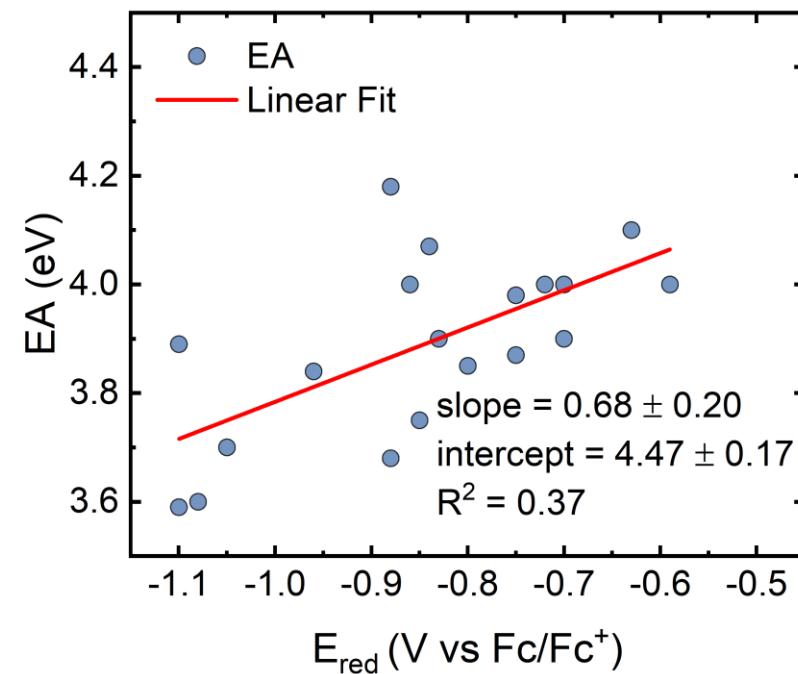
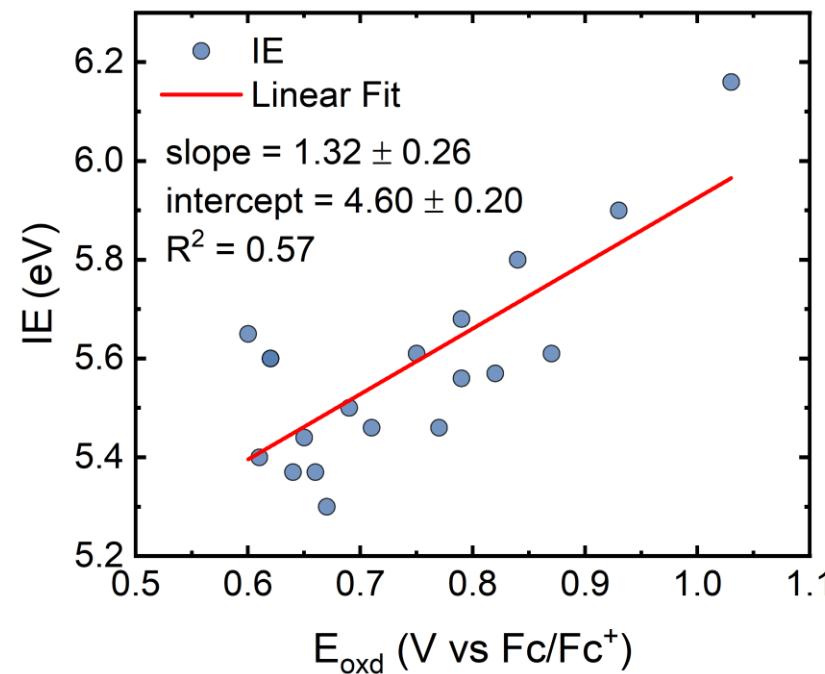
Donor Polymers





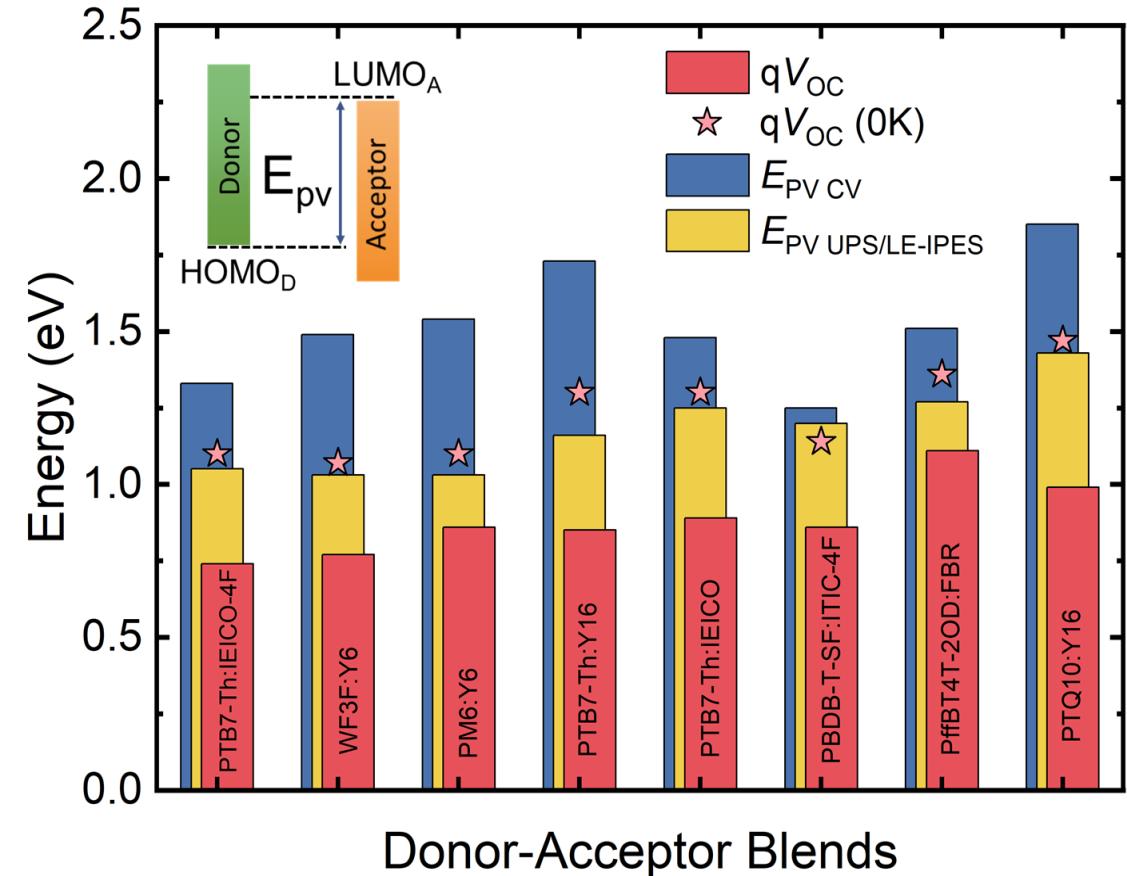
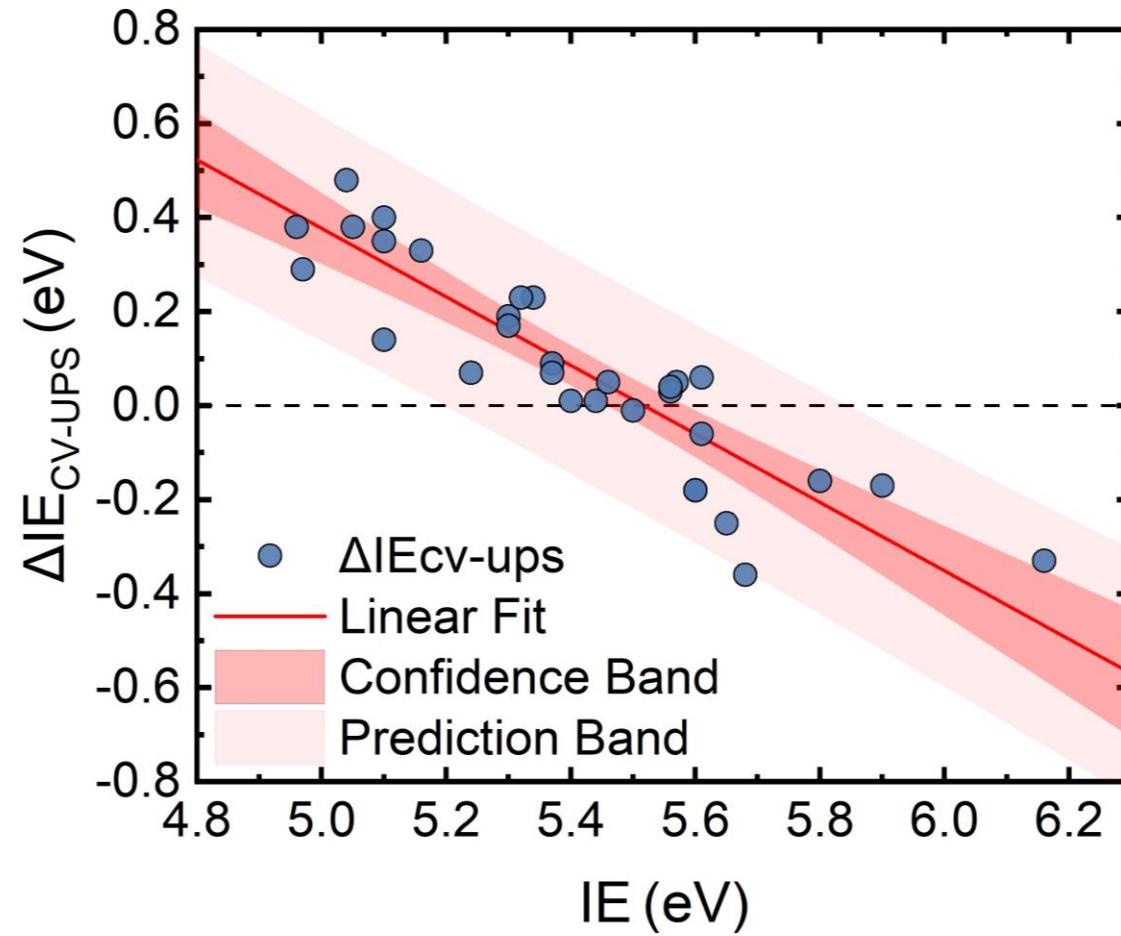
Correlating Redox Potentials with IE/EA

Non-fullerene Acceptors





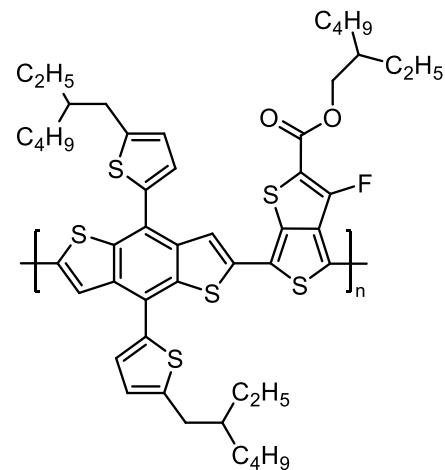
Correlating with IE/EA and Voc



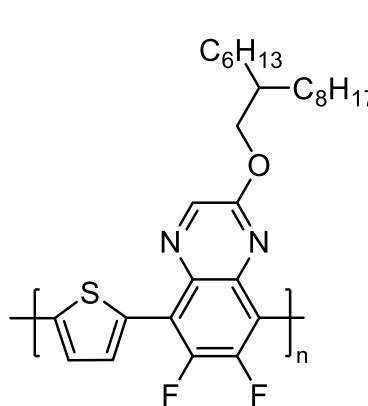


Role of IE-offset in Charge Generation

PTB7-Th



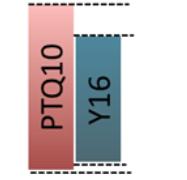
PTQ10



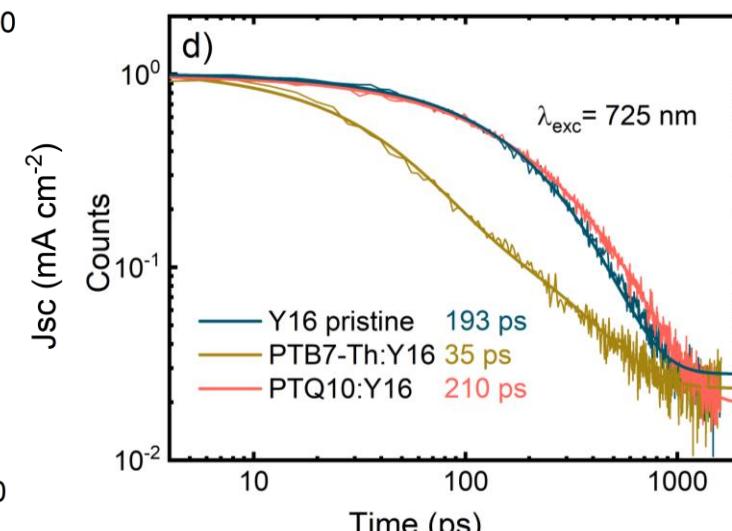
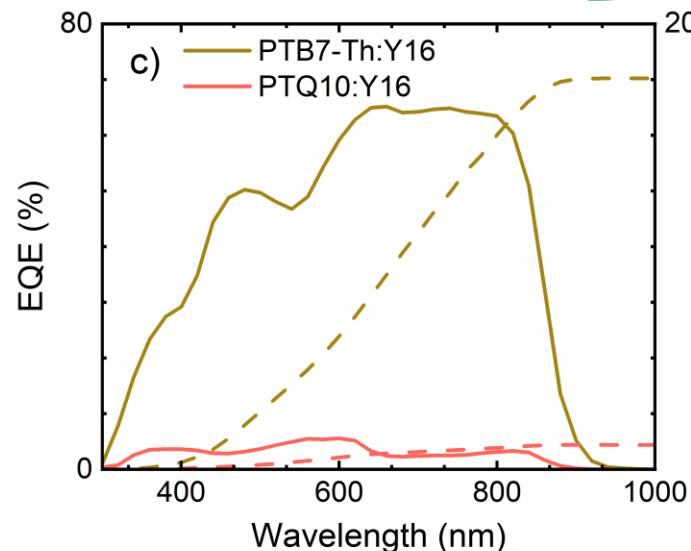
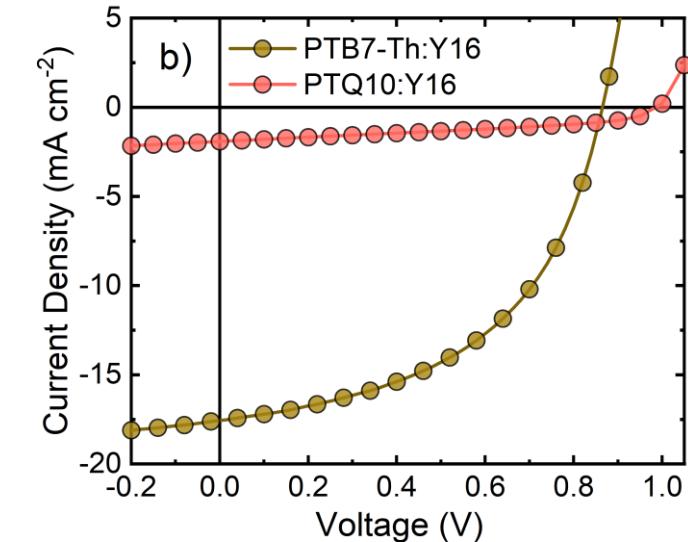
a) $\Delta_{E_{\text{red}}} = 0.04 \text{ eV}$



$\Delta_{E_{\text{red}}} = 0.72 \text{ eV}$

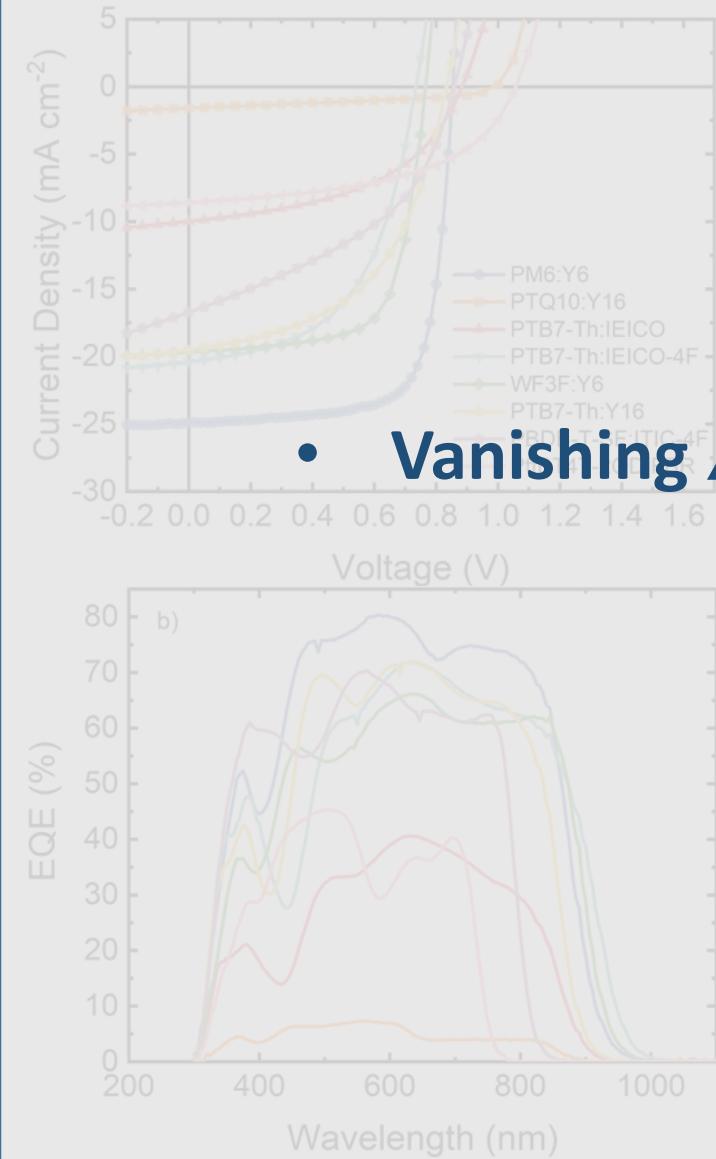


CV/LEIPES





Role of IE-offset in Charge Generation



BHJ Blends	ΔIE_{PES} (eV)	ΔIE_{CV} (eV)	J_{sc} (mA cm^{-2})	V_{oc} (V)	FF (%)	PCE (%)
PTQ10:Y16	0.05	0.09	1.59	0.99	41	0.64
PTQ10:IEICO-4F	0.18	0.06	2.81	0.82	41	0.25
PTQ10:COi8DFIC	0.28	0.13	7.4	0.83	42	0.95
PTB7-Th:Y16	0.31	0.03	19.5	0.87	51	2.56
PTB7-Th:Y1	0.39	0.02	16.6	0.79	40	8.3
PTB7-Th:IEICO-4F	0.45	0.06	20.4	0.74	54	5.3
WF3F:Y6	0.51	0.23	19.7	0.77	68	8.1
PM6:Y6	0.51	0.18	24.9	0.86	70	10.4

• **Vanishing ΔIE is DETERRIMENTAL For Device Performance**



جامعة الملك عبد الله
للغالوم والتكنولوجيا
King Abdullah University of
Science and Technology

KAUST
SOLAR
CENTER

Terpene Based Green Solvents for Stable Organic Photovoltaics



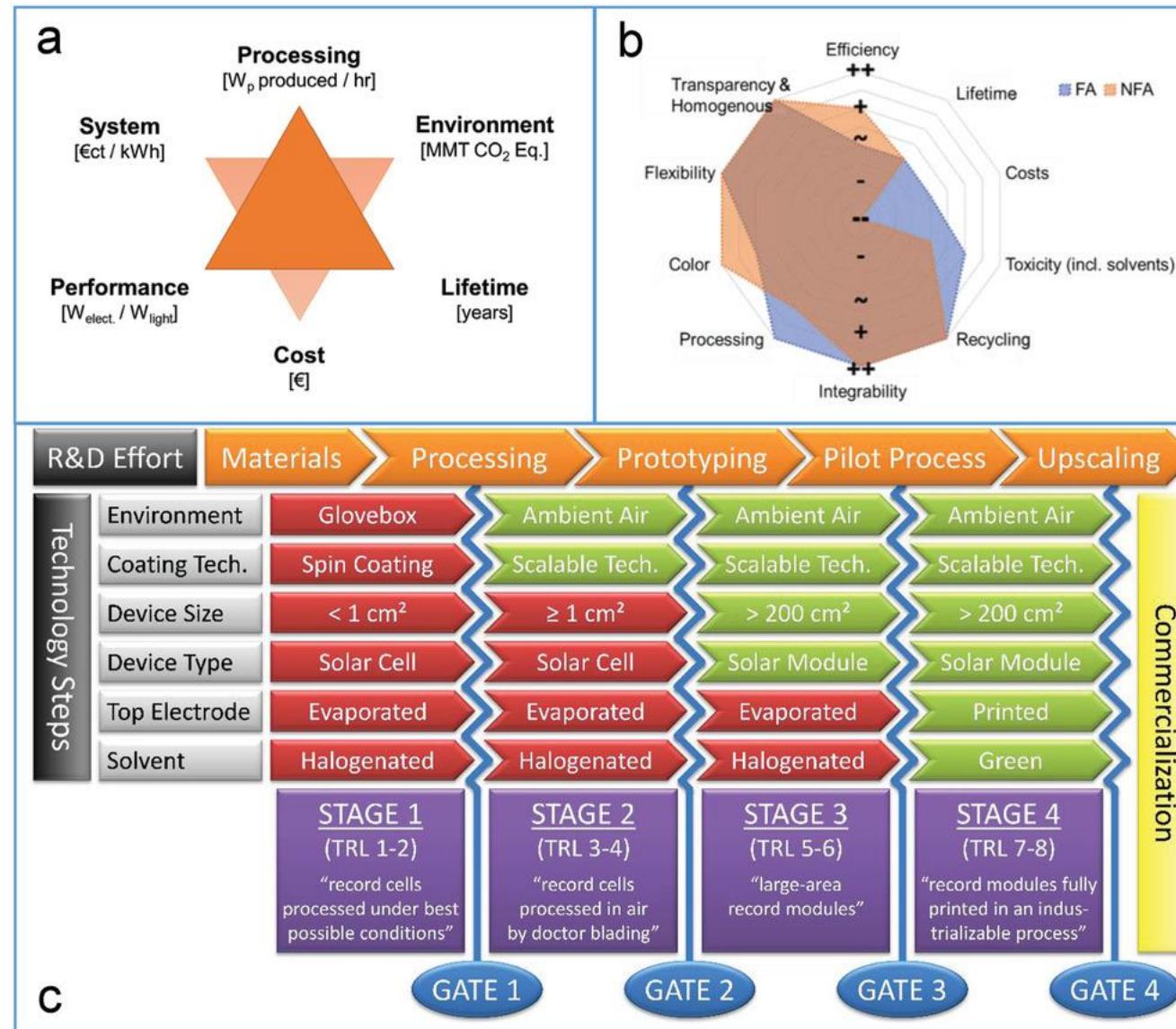
Daniel Corzo



Diego Rosas Villalva

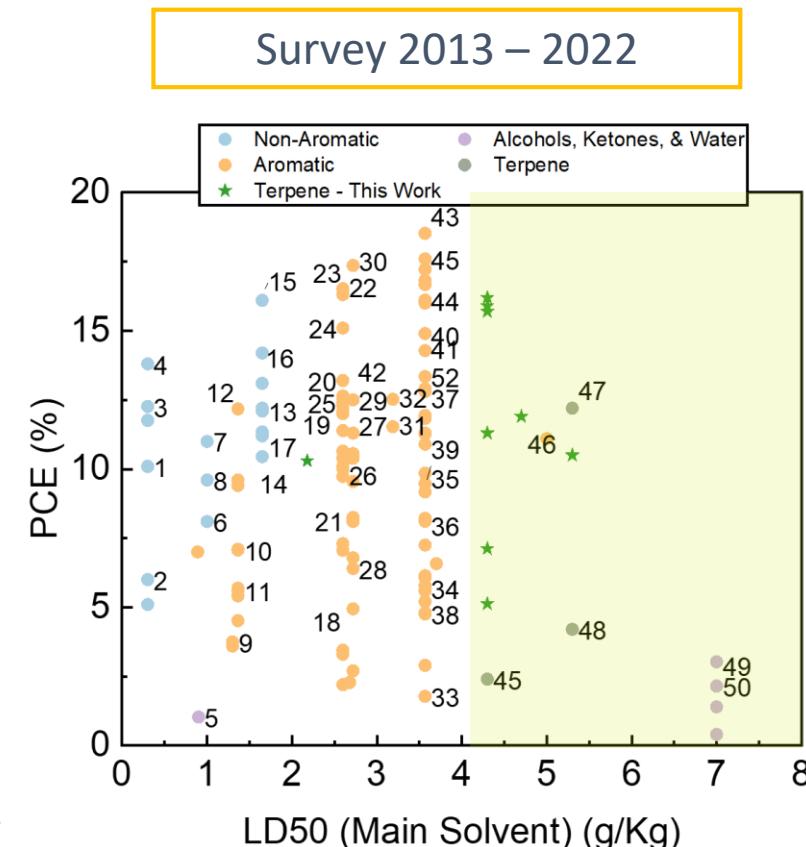
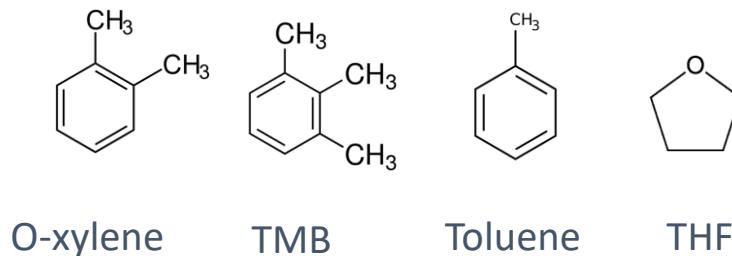
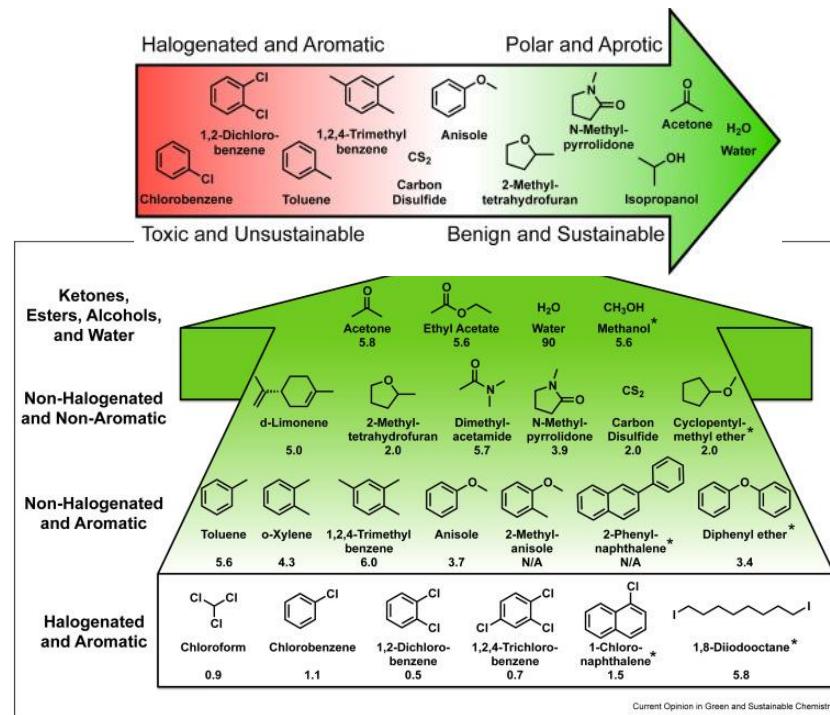


Opportunities for printed OPV

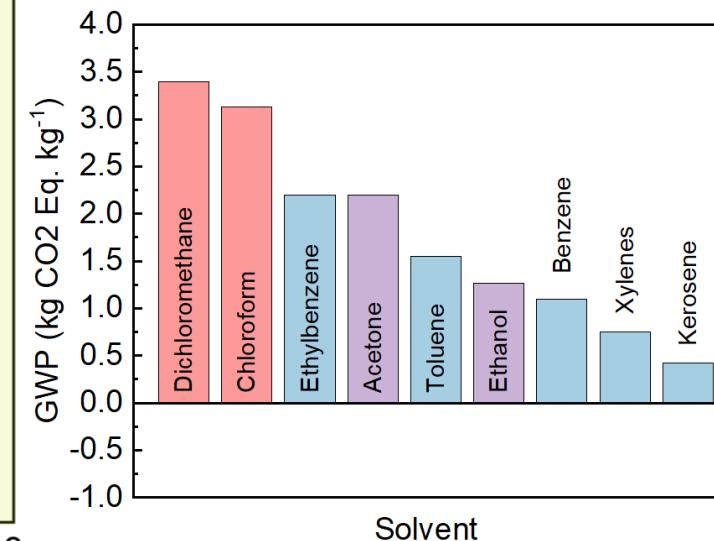


Strategies for green solvent transition

19



LD50 (Main Solvent) (g/Kg)
Lethal Dose: is one way to measure the short-term poisoning potential (acute toxicity) of a material.



Hansen solubility parameters

Hildebrandt Parameter

$$\delta_T = \sqrt{(E/V)}$$

$$\delta_T^2 = \delta_D^2 + \delta_P^2 + \delta_H^2$$

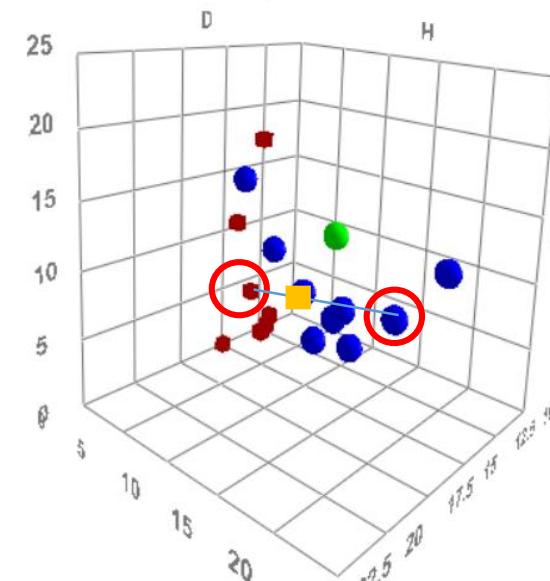
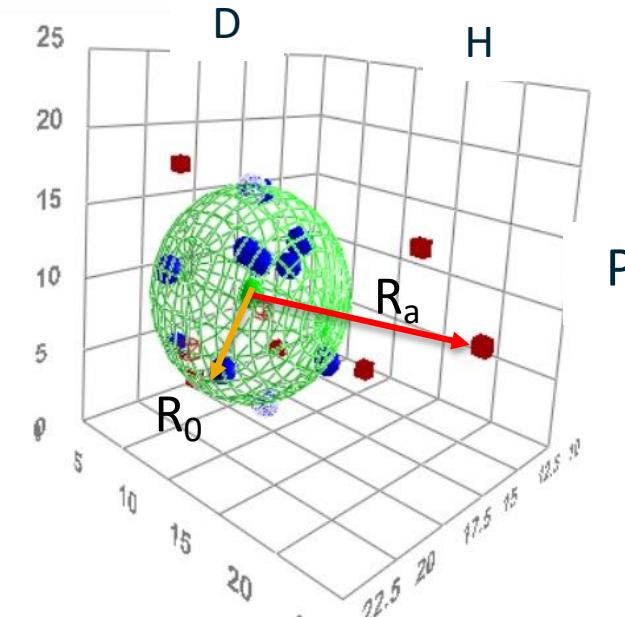
Dispersive Polar Hydrogen bonding

HSP distance - "alikeness"

$$(R_a)^2 = 4(\delta_{D1}-\delta_{D2})^2 + (\delta_{P1}-\delta_{P2})^2 + (\delta_{H1}-\delta_{H2})^2$$

HSP linear relationship (mixtures)

$$\delta_{blend} \equiv [\varphi_{comp1} \times \sigma_{com1}] + [\varphi_{comp2} \times \sigma_{com2}]$$



Relative Energy Distance = (R_a / R_0)

$RED < 1$ - Soluble

$RED = 1$ - P. Soluble

$RED > 1$ - N. Soluble

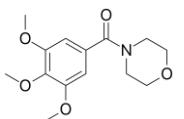
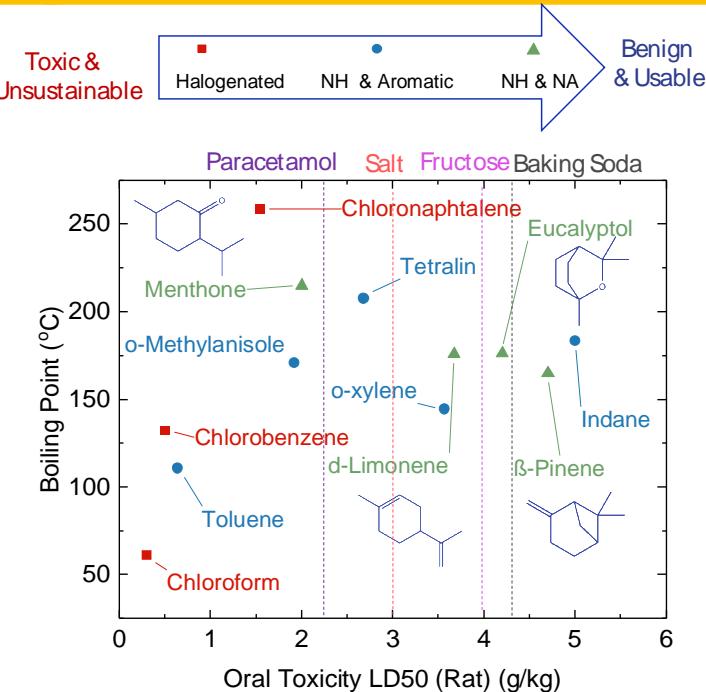


Finding Alternative Solvents

Formulation Guideline

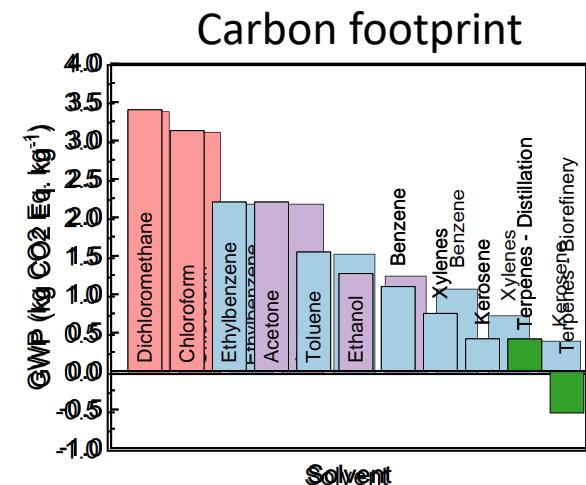
Terpenes as renewable solvents for OPV

21

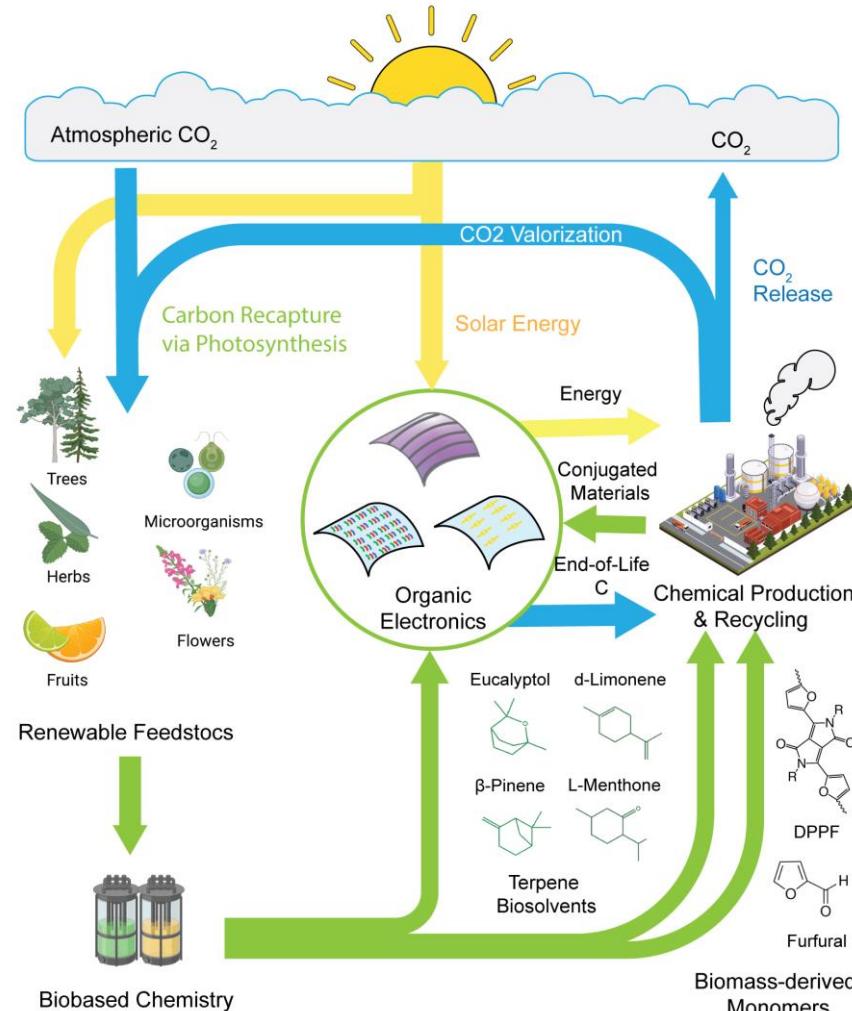


Drug Synthesis

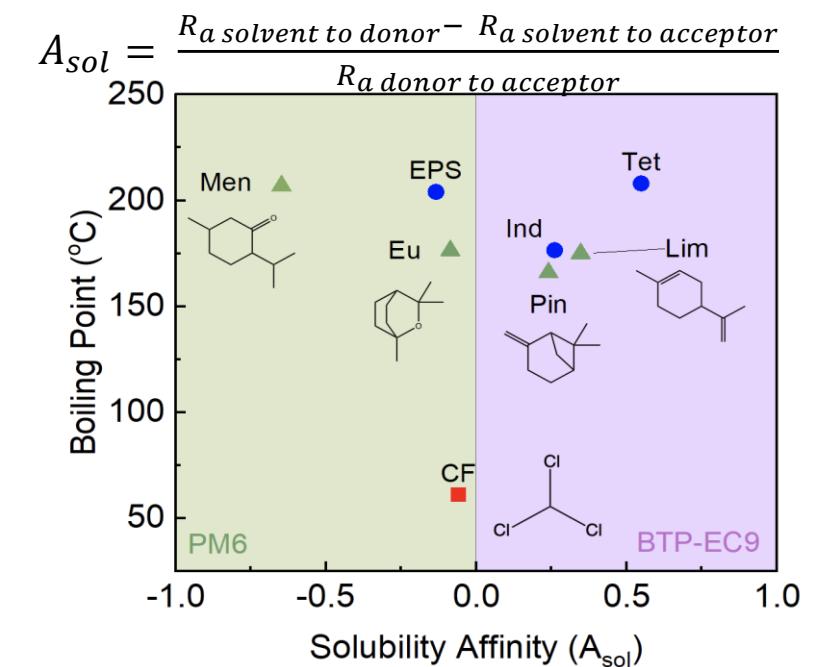
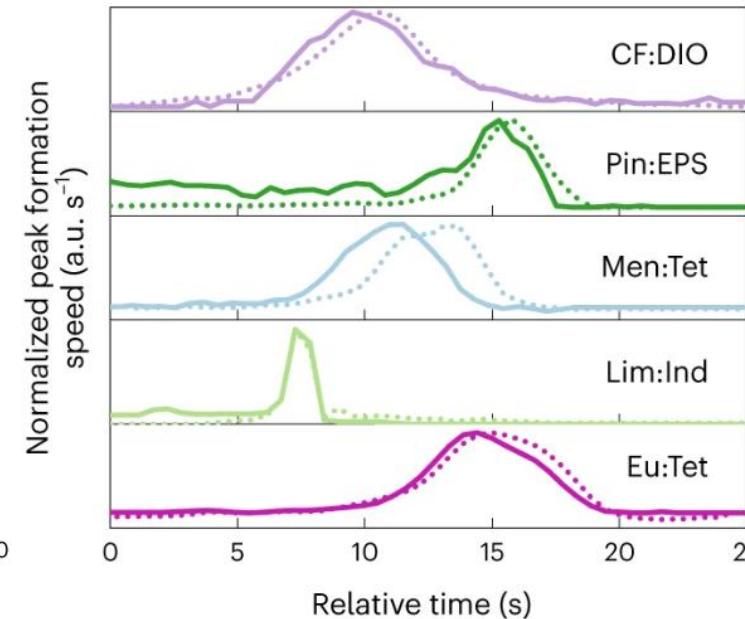
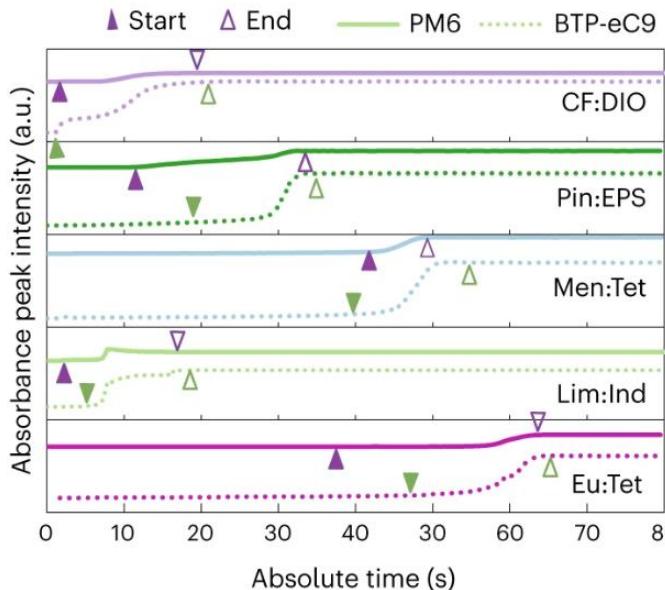
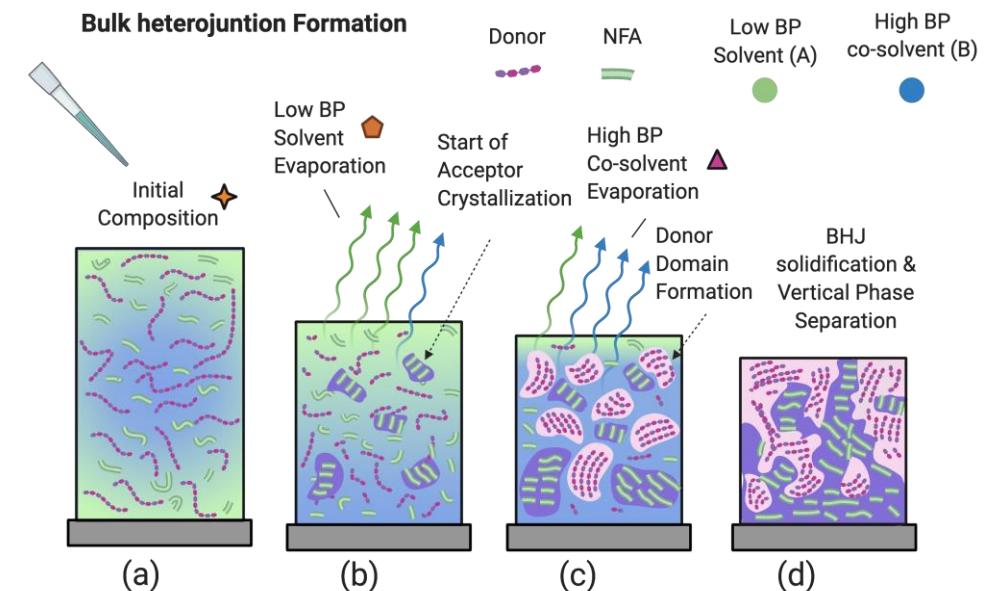
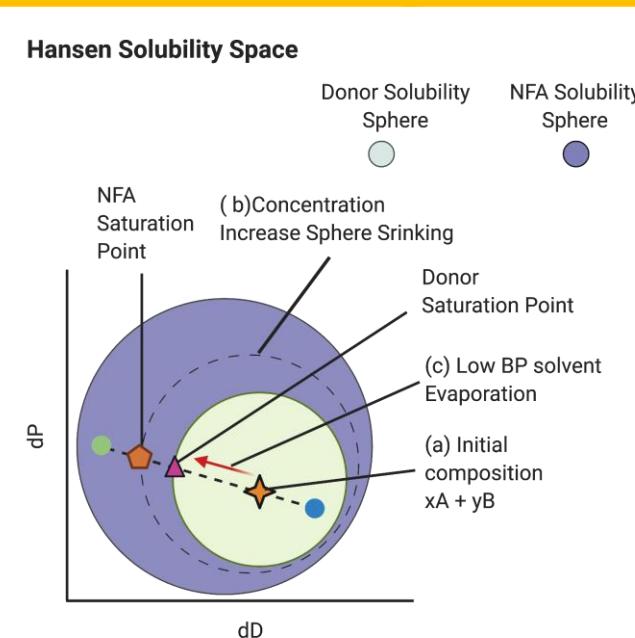
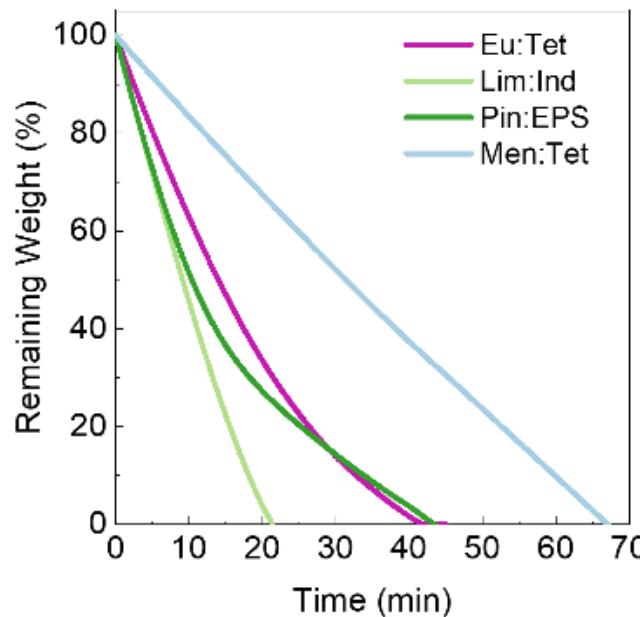
Trends in Biotechnology 35, 227-240
Org. Process Res. Dev. 2020, 24, 11, 2665–2675
Nature Reviews Materials 7, 117-137, (2022)



Ecoinvent database
Planta volume 249, pages 155–180 (2019)
Industrial & Engineering Chemistry Research 50, 11280-11287
Algal Research 23:1-11 (2017)
Chem. Commun., 2014, 50, 15288-15296



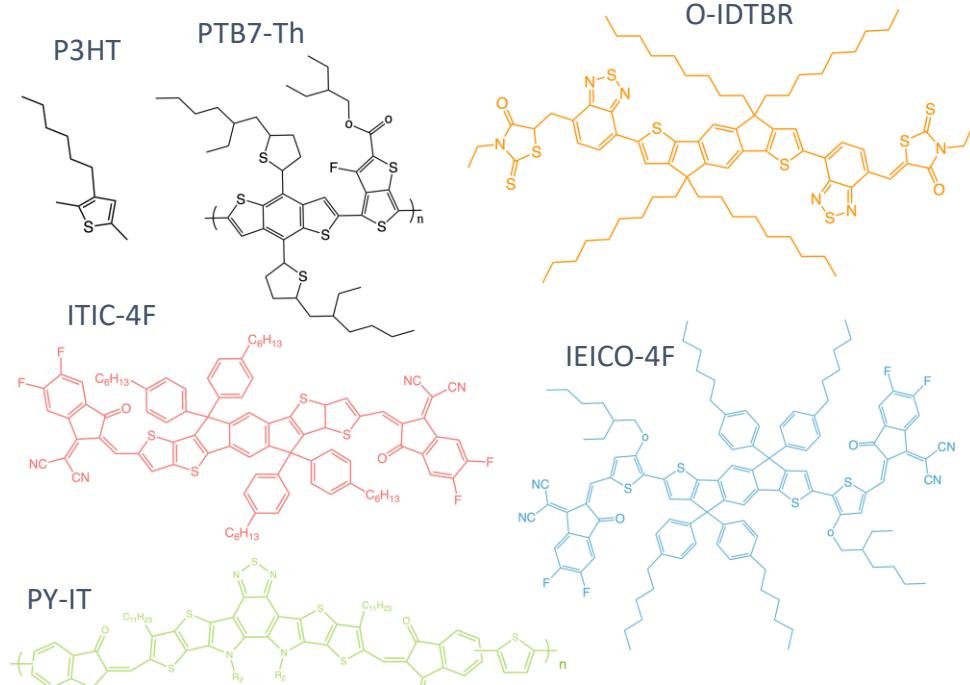
Ink formulation with terpenes



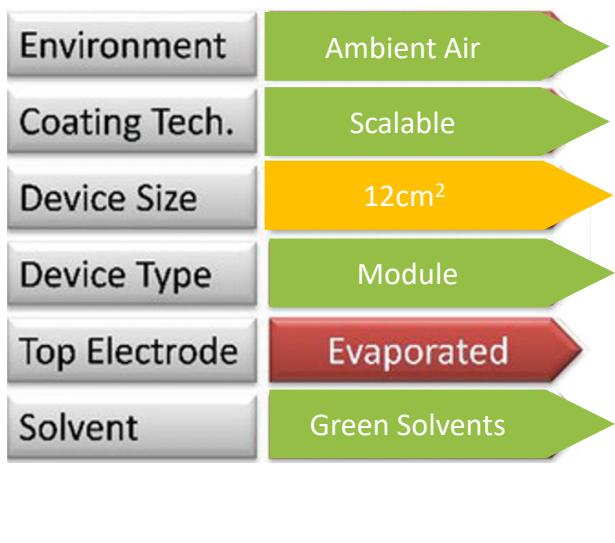
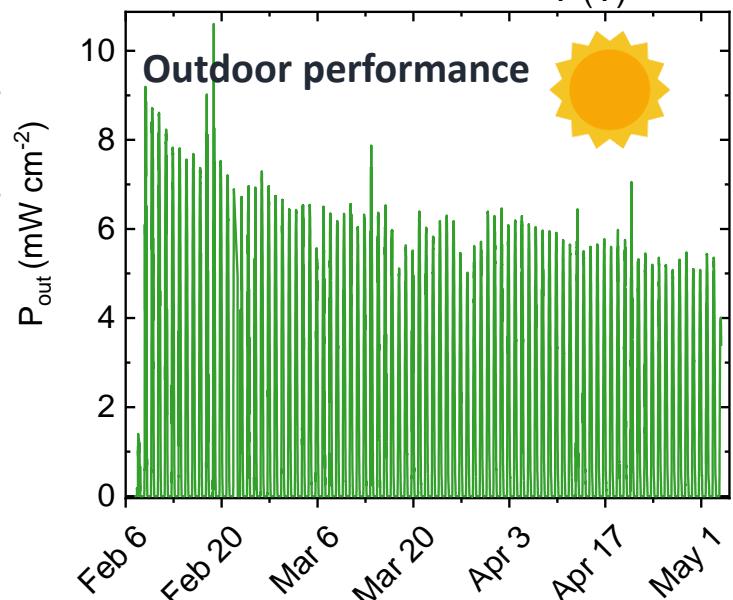
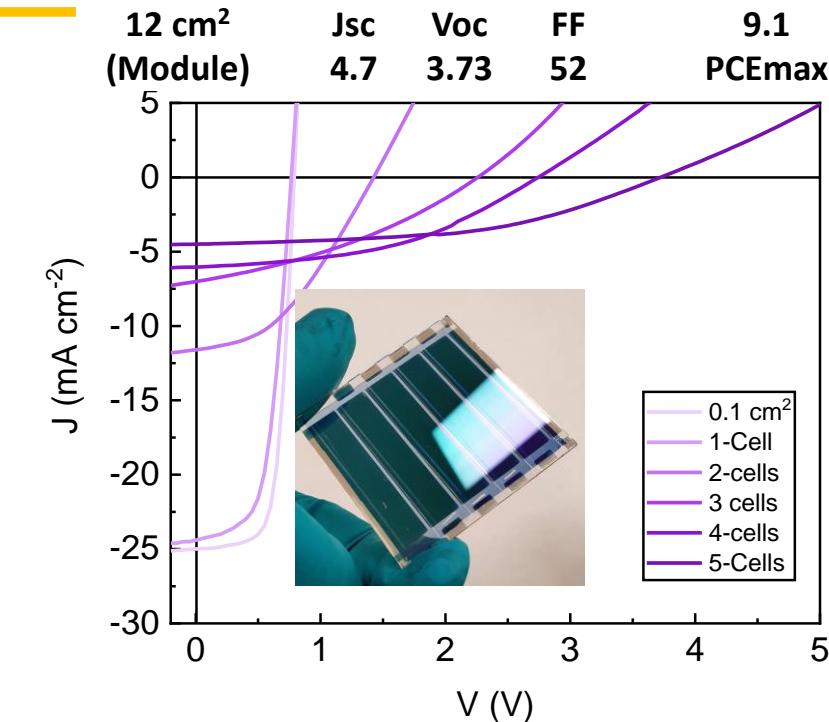
Terpene utilization in NFAs



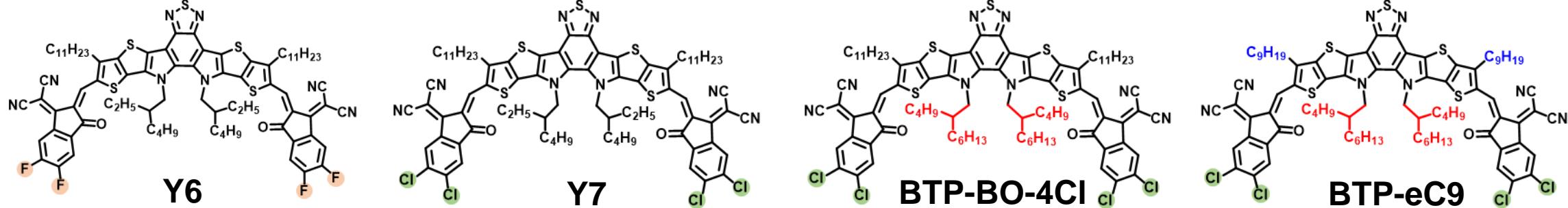
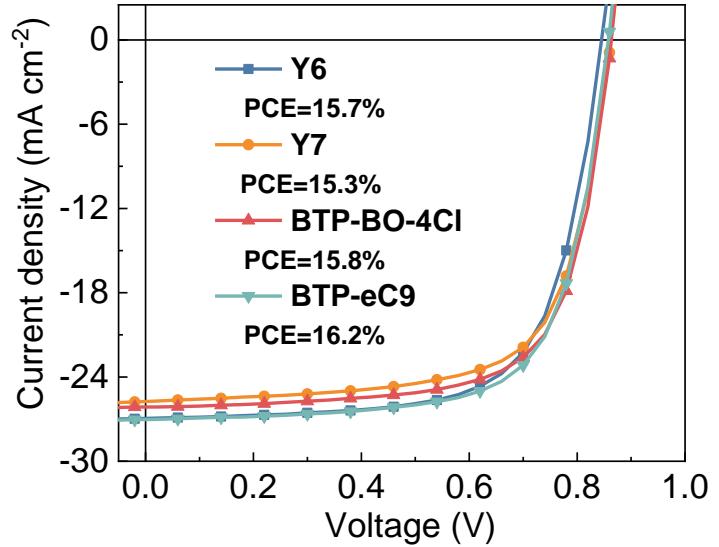
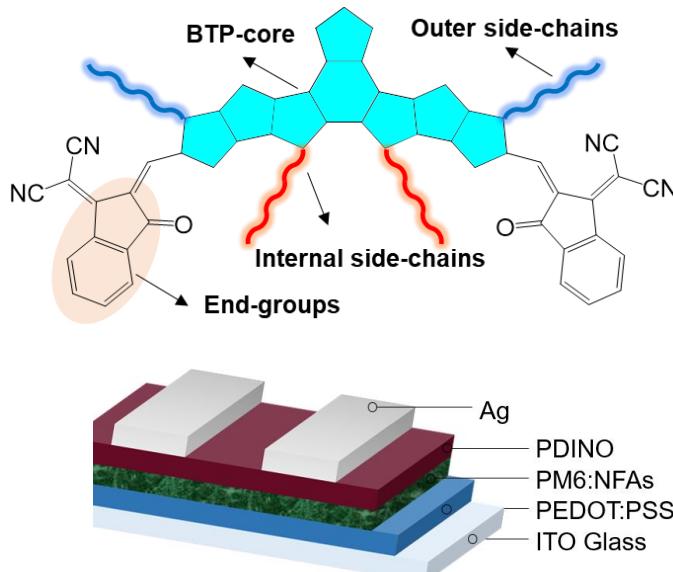
23



Blend	J_{sc} (mA cm $^{-2}$)	V_{oc} (V)	FF (%)	PCE_{ave} (%)	PCE_{max} (%)
P3HT:O-IDTBR (1:1)	11.5	0.72	60	5.1%	5.3%
PTB7-Th:IEICO-4F (1:1.5)	24.7	0.69	59	9.8%	10.6%
PM6:IT-4F (1:1.2)	15.2	0.81	55	6.8%	7.3%
PM6:PY-IT (1:1.2)	23.8	0.92	72	15.7%	15.9%
PM6:(PY-IT:BTP-eC9) (1:0.9:0.1)	25.5	0.89	71	16.1%	16.3%



Stability of Y-NFAs



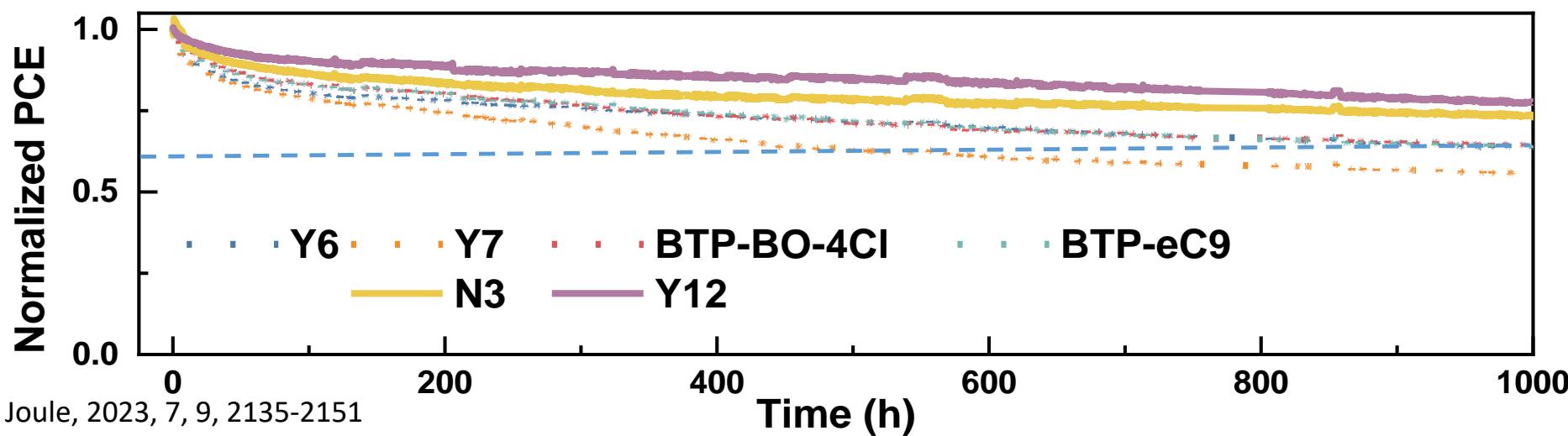
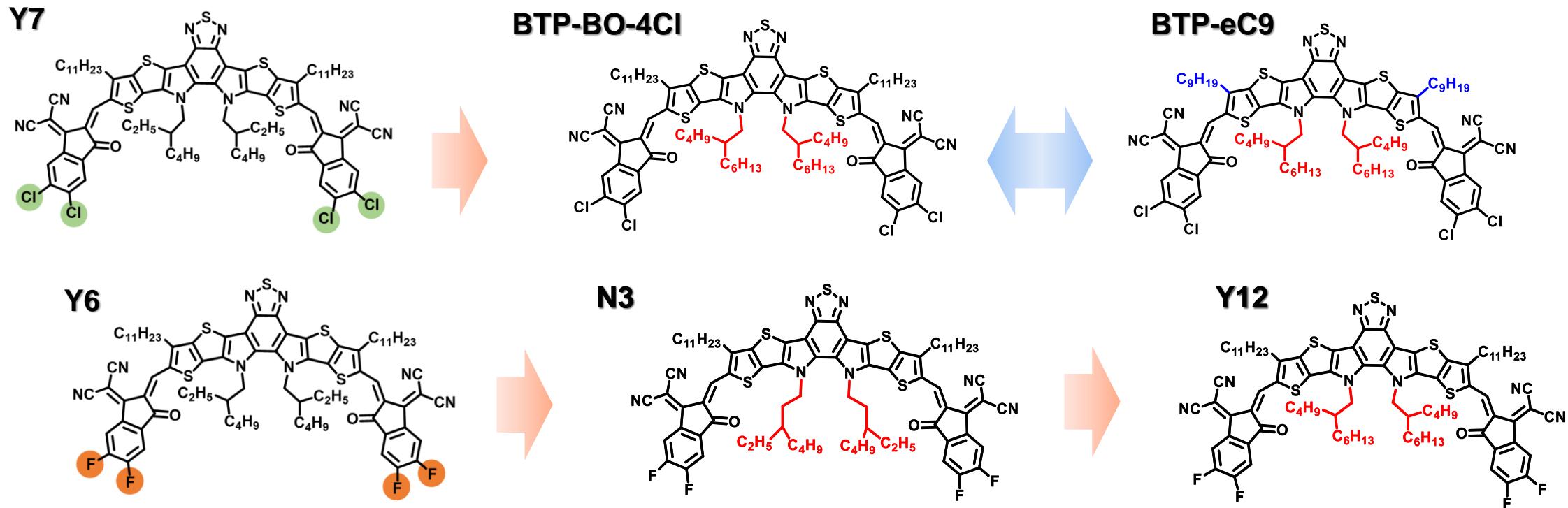
Optimized using No solvent additives

Table. Photovoltaic parameters of the devices

Acceptor ^a	J_{SC} (mA cm ⁻²)	V_{OC} (V)	FF (%)	PCE (%)
Y6	26.9	0.85	68.9	15.7
Y7	25.7	0.86	69.0	15.3
BTP-BO-4Cl	26.1	0.86	70.1	15.8
BTP-eC9	27.0	0.86	69.9	16.2

^a PM6:Y6=1.1.2, 100 °C annealing 10 min, w/o additive

Structure-stability relationship of Y-NFAs





THANK YOU FOR YOUR ATTENTION!

#EmbraceEquity
Globalyoungacademy.net



جامعة الملك عبد الله
للغعلوم والتكنولوجية
King Abdullah University of
Science and Technology

KAUST
SOLAR
CENTER